



Conceptual MCQs

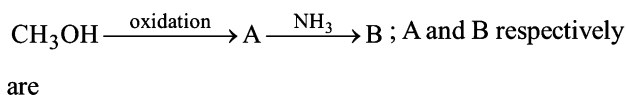
- Which one of the following alcohols is least soluble in water?
 - CH_3OH
 - $\text{C}_3\text{H}_7\text{OH}$
 - $\text{C}_4\text{H}_9\text{OH}$
 - $\text{C}_{10}\text{H}_{21}\text{OH}$
- How many isomers of $\text{C}_5\text{H}_{11}\text{OH}$ will be primary alcohols?
 - 5
 - 4
 - 2
 - 3
- Ethyl alcohol can be prepared from Grignard reagent by the reaction of:
 - HCHO
 - R_2CO
 - RCN
 - RCOI
- Methylated spirit is:
 - methanol
 - methanol + ethanol
 - methanoic acid
 - methanamide
- Glycerol is more viscous than ethanol due to:
 - high molecular weight.
 - high boiling point.
 - many hydrogen bonds per molecule.
 - Fajan's rule.
- Carbolic acid is:
 - phenol
 - phenyl benzoate
 - phenyl acetate
 - salol
- Which one of the following compounds will be most readily attacked by an electrophile?
 - Chlorobenzene
 - Benzene
 - Phenol
 - Toluene
- The IUPAC name of $\text{CH}_3 - \underset{\text{OH}}{\text{CH}} - \text{CH}_2 - \overset{\text{CH}_3}{\underset{\text{OH}}{\text{C}}} - \text{CH}_3$ is:
 - 1, 1-dimethyl-1, 3-butanediol
 - 2-methyl-2, 4-pentanediol
 - 4-methyl-2, 4-pentanediol
 - 1, 3, 3-trimethyl-1, 3-propanediol
- Which of the following has lowest boiling point?
 - p*-Nitrophenol
 - m*-Nitrophenol
 - o*-Nitrophenol
 - Phenol
- HBr reacts fastest with:
 - 2-Methylpropan-1-ol
 - 2-Methylpropan-2-ol
 - propan-2-ol
 - propan-1-ol.
- Williamson's synthesis is used to prepare
 - acetone
 - diethyl ether
 - P.V.C.
 - bakelite
- What is formed when a primary alcohol undergoes catalytic dehydrogenation?
 - Aldehyde
 - Ketone
 - Alkene
 - Acid
- The reaction of sodium ethoxide with ethyl iodide to form diethyl ether is termed:
 - electrophilic substitution
 - nucleophilic substitution
 - electrophilic addition
 - radical substitution
- Which one of the following is not formed when glycerol reacts with HI ?
 - $\text{CH}_3 - \text{CHI} - \text{CH}_3$
 - $\text{CH}_3 - \text{CH} = \text{CH}_2$
 - $\text{CH}_2\text{OH} - \text{CHI} - \text{CH}_2\text{OH}$
 - $\text{CH}_2 = \text{CH} - \text{CH}_2\text{I}$
- Which of the following product is formed, when ether is exposed to air?
 - Oxide
 - Alkanes
 - Alkene
 - Peroxide of diethyl ether



Application Based MCQs

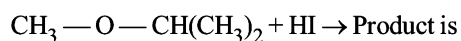
16. Which is formed when benzylamine react with nitrous acid?
 (a) C_6H_5OH (b) C_6H_5ON
 (c) $C_6H_5N_2OH$ (d) $C_6H_5CH_2OH$
17. *n*-Propyl alcohol and isopropyl alcohol can be chemically distinguished by which reagent?
 (a) PCl_5
 (b) Reduction
 (c) Oxidation with potassium dichromate
 (d) Ozonolysis
18. The major product obtained on interaction of phenol with sodium hydroxide and carbon dioxide is :
 (a) salicylaldehyde (b) salicylic acid
 (c) phthalic acid (d) benzoic acid
19. Ethylene oxide when treated with Grignard reagent yields :
 (a) tertiary alcohol (b) cyclopropyl alcohol
 (c) primary alcohol (d) secondary alcohol
20. From amongst the following alcohols the one that would react fastest with conc. HCl and anhydrous $ZnCl_2$, is :
 (a) 2-Butanol (b) 2-Methylpropan-2-ol
 (c) 2-Methylpropanol (d) 1-Butanol
21. Among the following the one that gives positive iodoform test upon reaction with I_2 and NaOH is :
 (a) $CH_3 - \overset{\overset{CH_2}{|}}{CH}CH_2OH$
 (b) $PhCHOHCH_3$
 (c) $CH_3CH_2CH(OH)CH_2CH_3$
 (d) $C_6H_5CH_2CH_2OH$
22. The main product of the following reaction is
 $C_6H_5CH_2CH(OH)CH(CH_3)_2 \xrightarrow{\text{conc. } H_2SO_4} ?$
 (a) $H_5C_6 - \overset{\overset{H}{|}}{C} = C - \overset{\overset{H}{|}}{CH(CH_3)_2}$
 (b) $C_6H_5CH_2 - \overset{\overset{H}{|}}{C} = C - \overset{\overset{CH_3}{|}}{CH_3}$
 (c) $H_5C_6CH_2CH_2 - \overset{\overset{H_3C}{|}}{C} = CH_2$
 (d) $C_6H_5 - \overset{\overset{H}{|}}{C} = C - \overset{\overset{CH(CH_3)_2}{|}}{H}$
23. The best method to prepare cyclohexene from cyclohexanol is by using :
 (a) Conc. HCl + $ZnCl_2$ (b) Conc. H_3PO_4
 (c) HBr (d) Conc. HCl
24. The correct order of acid strength of the following compounds :
 (A) Phenol (B) *p*-Cresol
 (C) *m*-Nitrophenol (D) *p*-Nitrophenol
 (a) $D > C > A > B$ (b) $B > D > A > C$
 (c) $A > B > D > C$ (d) $C > B > A > D$
25. In order to get 2-hydroxybenzaldehyde from phenol, which of the following reagents is required ?
 (a) $(CH_3CO)_2O, H_2SO_4$ (b) $CHCl_3/NaOH$
 (c) $CO_2, NaOH$ (d) $CCl_4/NaOH$
26. Consider the following reaction :
 $C_2H_5OH + H_2SO_4 \longrightarrow \text{Product}$
 Among the following, which one cannot be formed as a product under any conditions ?
 (a) Ethylene (b) Acetylene
 (c) Diethyl ether (d) Ethyl-hydrogen sulphate
27. Dehydration of 2-butanol yields :
 (a) 1-butene (b) 2-butene
 (c) 2-butyne (d) both (a) and (b)
28. Phenol is heated with a solution of mixture of KBr and $KBrO_3$. The major product obtained in the above reaction is :
 (a) 2-Bromophenol (b) 3-Bromophenol
 (c) 4-Bromophenol (d) 2,4,6-Tribromophenol
29. The dehydration of 2-methylbutanol with conc. H_2SO_4 gives:
 (a) 2-Methylbutene as major product
 (b) Pentene
 (c) 2-Methylbut-2-ene as major product
 (d) 2-Methylpent-2-ene
30. Which of the following reagents may be used to distinguish between phenol and benzoic acid?
 (a) Aqueous NaOH (b) Tollen's reagent
 (c) Molisch reagent (d) Neutral $FeCl_3$

31. In the reaction :



- (a) $\text{HCHO}, \text{HCOONH}_4$ (b) $\text{HCOOH}, \text{HCOONH}_4$
 (c) $\text{HCOOH}, \text{HCONH}_2$ (d) $\text{HCHO}, \text{HCONH}_2$

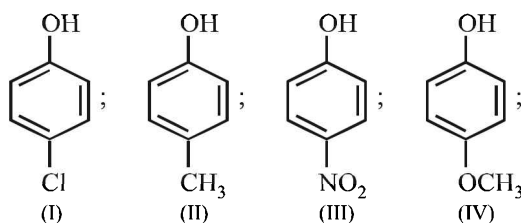
32. The major organic product in the reaction,



- (a) $\text{ICH}_2\text{OCH}(\text{CH}_3)_2$
 (b) $\text{CH}_3\text{OC}(\text{CH}_3)_2$
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- (c) $\text{CH}_3\text{I} + (\text{CH}_3)_2\text{CHOH}$
 (d) $\text{CH}_3\text{OH} + (\text{CH}_3)_2\text{CHI}$

33. Arrange the following compounds in order of decreasing acidity :



- (a) $\text{II} > \text{IV} > \text{I} > \text{III}$ (b) $\text{I} > \text{II} > \text{III} > \text{IV}$
 (c) $\text{III} > \text{I} > \text{II} > \text{IV}$ (d) $\text{IV} > \text{III} > \text{I} > \text{II}$

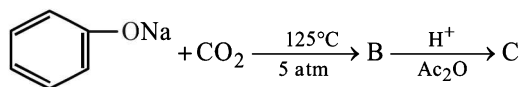
34. The most suitable reagent for the conversion of $\text{R} - \text{CH}_2 - \text{OH} \longrightarrow \text{R} - \text{CHO}$ is:

- (a) KMnO_4
 (b) $\text{K}_2\text{Cr}_2\text{O}_7$
 (c) CrO_3
 (d) PCC (Pyridinium chlorochromate)

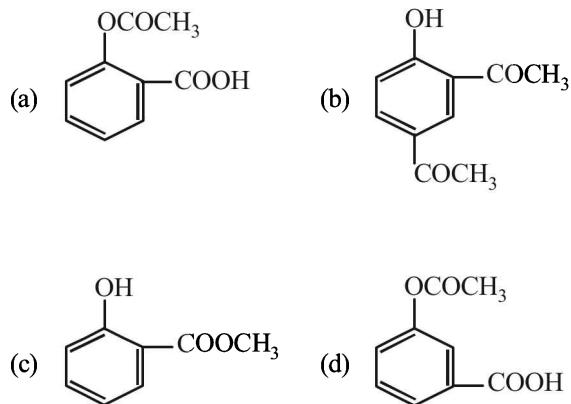
35. Which of the following reagent is best to change glycerol to acrolein ?

- (a) P_2O_5 (b) Conc. H_2SO_4
 (c) Anhydrous CaCl_2 (d) KHSO_4

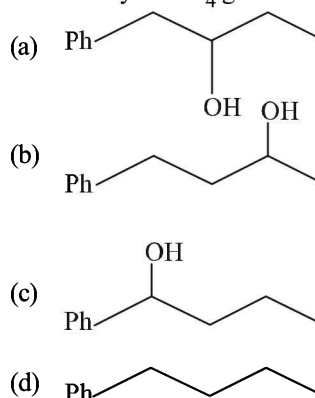
36. Sodium phenoxide when heated with CO_2 under pressure at 125°C yields a product which on acetylation produces C



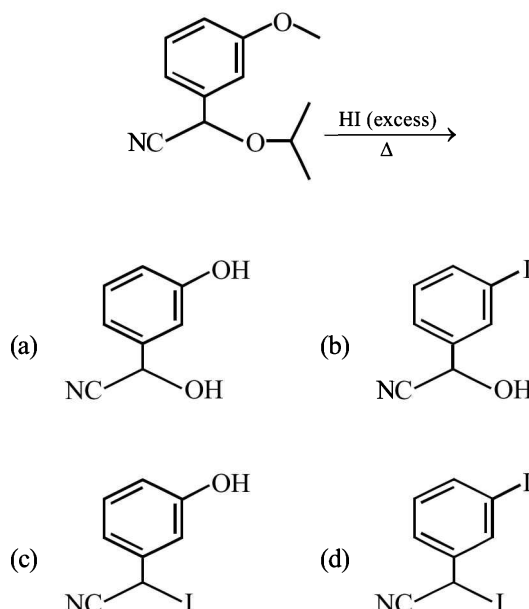
The major product C would be :



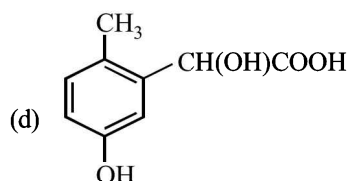
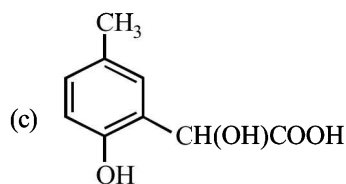
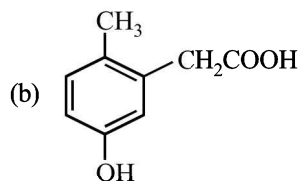
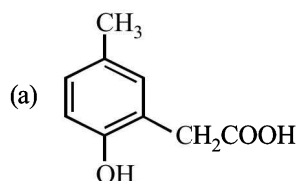
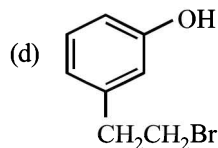
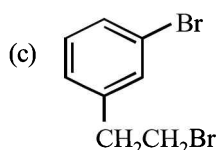
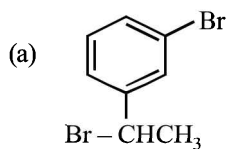
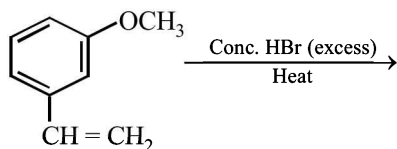
37. Heating of 2-chloro-1-phenylbutane with EtOK/EtOH gives X as the major product. Reaction of X with $\text{Hg}(\text{OAc})_2/\text{H}_2\text{O}$ followed by NaBH_4 gives Y as the major product. Y is :



38. The major product of the following reaction is:



39. The major product of the following reaction is :

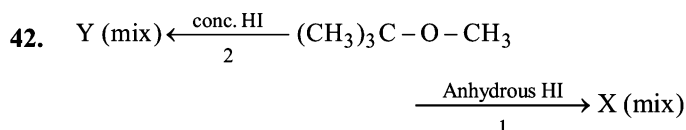
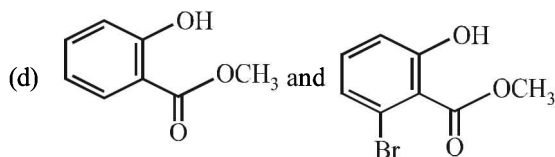
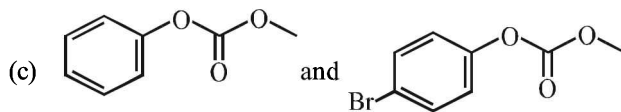
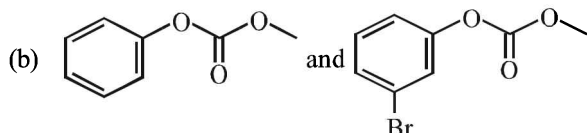
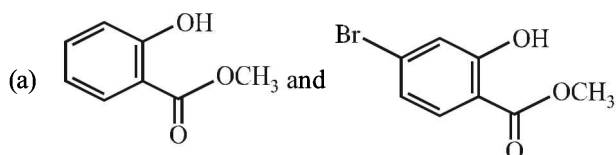


40. *p*-cresol reacts with chloroform in alkaline medium to give the compound A which adds hydrogen cyanide to form, the compound B. The latter on acidic hydrolysis gives chiral carboxylic acid. The structure of the carboxylic acid is



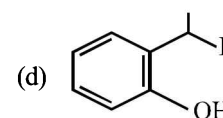
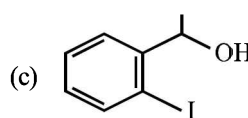
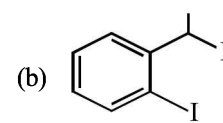
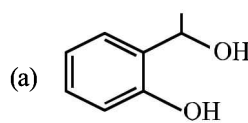
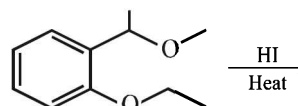
Skill Based MCQs

41. Phenol reacts with methyl chloroformate in the presence of NaOH to form product A. A reacts with Br_2 to form product B. A and B are respectively :

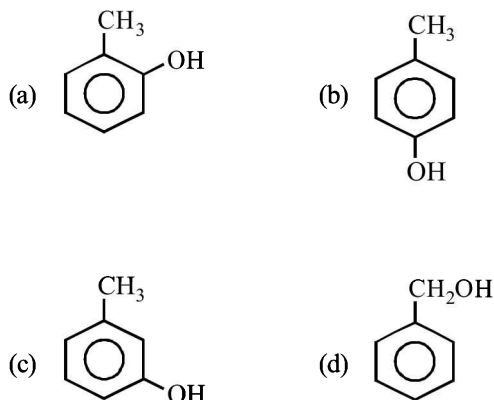


- (a) X and Y are identical mixture of CH_3I and $(\text{CH}_3)_3\text{C-OH}$
 (b) X and Y are identical mixture of CH_3OH & $(\text{CH}_3)_3\text{C-I}$
 (c) X is mixture of CH_3I and $(\text{CH}_3)_3\text{C-OH}$
 (d) Y is mixture of CH_3OH & $(\text{CH}_3)_3\text{C-I}$

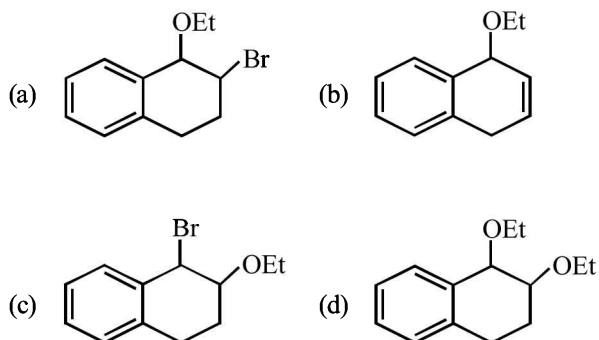
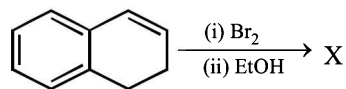
43. The major product formed in the following reaction is :



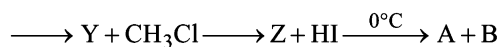
44. The structure of the compound that gives a tribromo derivative on treatment with bromine water is



45. The major product (X) of the following reaction is:



46. $\text{CH}_3 - \text{CH} = \text{CH}_2 \xrightarrow[\text{(ii) NaBH}_4]{\text{(i) Hg(OAc)}_2/\text{H}_2\text{O}} \text{X} + \text{Na}$



What are A and B ?

- (a) $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{OH}$ & CH_3I
- (b) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \text{OH}$ & CH_3I
- (c) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \text{I}$ & CH_3OH
- (d) $\text{CH}_3 - \text{CH}_2\text{CH}_2\text{I}$ & CH_3OH

47. $\text{CH}_3\underset{\text{CH}_3}{\text{CH}} - \text{CH} = \text{CH}_2 \xrightarrow[\text{(ii) H}_2\text{O}_2/\text{OH}^-]{\text{(i) B}_2\text{H}_6} \text{X} \xrightarrow[140^\circ\text{C}]{\text{H}_2\text{SO}_4} \text{Y}$

What is Y ?

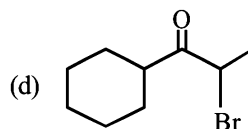
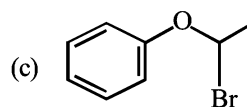
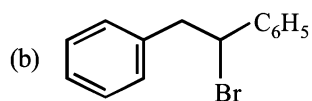
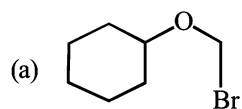
- (a) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_2 - \text{CH}_2 - \text{O} - \text{CH}_2 - \text{CH}_2 - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_3$
- (b) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \text{CH} = \text{CH}_2$
- (c) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \underset{\text{CH}_3}{\text{CH}} - \text{O} - \underset{\text{CH}_3}{\text{CH}} - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_3$
- (d) $\text{CH}_3 - \underset{\text{C}_2\text{H}_5}{\underset{\text{CH}_3}{\text{C}}} - \text{O} - \underset{\text{C}_2\text{H}_5}{\underset{\text{CH}_3}{\text{C}}} - \text{CH}_3$

48. $(\text{X}) \xrightarrow[\text{(C}_5\text{H}_{10}\text{O)}]{\text{H}_3\text{O}^+} \text{Y} + \text{Z}$

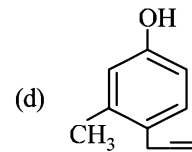
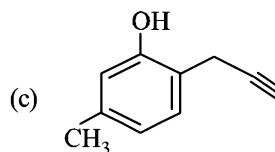
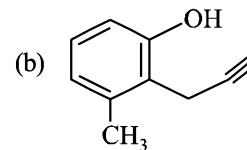
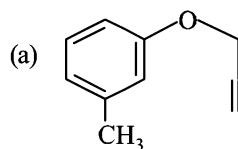
(Y and Z both give the Iodoform test). The compound X is—

- (a) $\text{CH}_3 - \text{CH} = \text{CH} - \text{O} - \text{CH}_2 - \text{CH}_3$
- (b) $\text{CH}_3 - \underset{\text{CH}_3}{\overset{\text{H}}{\text{C}}} - \text{O} - \text{CH}_2 - \text{CH}_3$
- (c) $\text{CH}_3 - \underset{\text{CH}_2}{\text{C}} = \text{O} - \text{CH}_2 - \text{CH}_3$
- (d) Both (a) and (c)

49. Which of the following, upon treatment with *tert*-BuONa followed by addition of bromine water, fails to decolourize the colour of bromine?



50. What will be the major product when *m*-cresol is reacted with propargyl bromide ($\text{HC} \equiv \text{C} - \text{CH}_2\text{Br}$) in presence of K_2CO_3 in acetone?



ANSWER KEY

Conceptual MCQs

1	(d)	3	(a)	5	(c)	7	(c)	9	(c)	11	(b)	13	(b)	15	(d)				
2	(b)	4	(b)	6	(a)	8	(b)	10	(b)	12	(a)	14	(c)						

Application Based MCQs

16	(d)	19	(c)	22	(a)	25	(b)	28	(d)	31	(b)	34	(d)	37	(c)	40	(c)		
17	(c)	20	(b)	23	(b)	26	(b)	29	(c)	32	(c)	35	(d)	38	(c)				
18	(b)	21	(b)	24	(a)	27	(d)	30	(d)	33	(c)	36	(a)	39	(b)				

Skill Based MCQs

41	(c)	42	(b)	43	(d)	44	(c)	45	(a)	46	(b)	47	(a)	48	(c)	49	(a)	50	(a)
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