

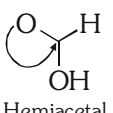
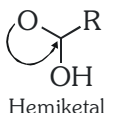
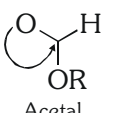
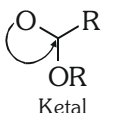
CARBOHYDRATES

- Carbohydrates are defined as optically active polyhydroxy aldehydes or ketones or the compound which produce such units on hydrolysis.
- Monosaccharide** ($C_nH_{2n}O_n$) : single unit, can't be hydrolysed : Glucose and fructose.
- Oligosaccharides** gives two to ten monosaccharides on hydrolysis.
- Disaccharides** (by glycosidic linkage)

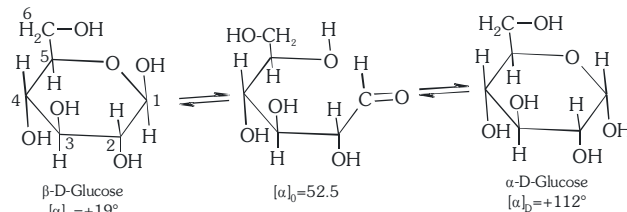
Sucrose $\xrightarrow{H_3O^+}$ α -D. Glucose + β -D. Fructose;

Maltose $\xrightarrow{H_3O^+}$ 2 α -D. Glucose unit

Lactose $\xrightarrow{H_3O^+}$ β -D. Glucose + β -D. Galactose
- Polysaccharide** : Contain more than ten monosaccharide units
 $(C_6H_{10}O_5)_n$: Starch & cellulose.

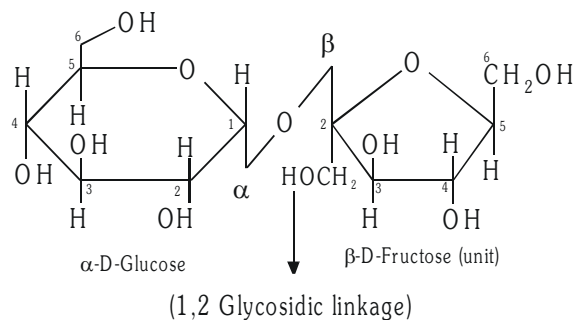
TYPE OF SUGAR		
Give Test	Reducing	Non Reducing
Tollen's Reagent	+ve test	-ve test
Fehling Reagent	+ve test	-ve test
Benedict Test	+ve test	-ve test
Mutarotation	Yes	No
Functional Unit	$-\overset{\alpha}{C}-C=O / -\overset{O}{\underset{OH}{\underset{ }{C}}}-C-$  Hemiacetal  Hemiketal	 Acetal  Ketal
Example	All monosaccharides Glucose Fructose Mannose Galactose Disaccharide Maltose lactose	Disaccharide Sucrose Polysaccharide Starch cellulose

- Mutarotation**: When either form of D-glucose is placed in aq. solution it slowly form the other via open chain aldehyde and gradual change in specific rotation until specific rotation ($\pm 52.5^\circ$) is reached.

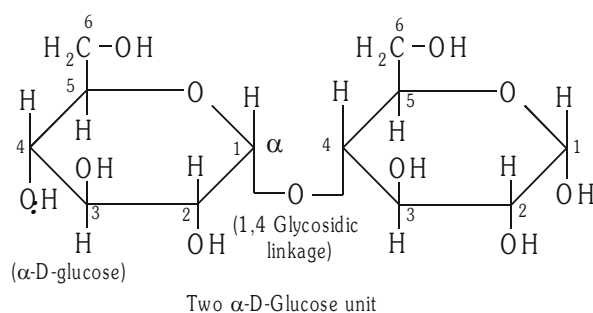


- Anomer's** : Differ in configuration at 1st carbon due to hemi (acetal or ketal) ring formation. The new-asymmetric carbon is referred to as Anomeric carbon.
- Epimer's** : Diastereomer's which differ in conformation at any one chiral carbon
 eg. D-Glucose & D-mannose
 D-Glucose & D-Galactose

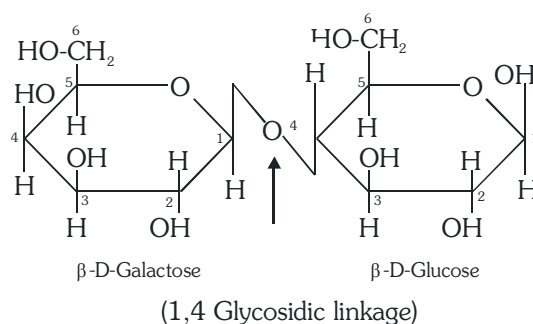
- Sucrose** :



- Maltose**

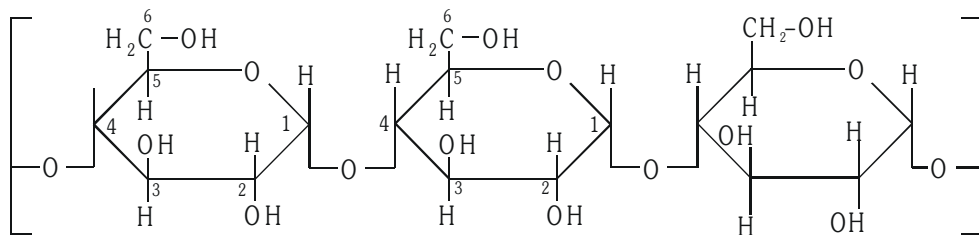


- Lactose** :



- **Starch** : (Amylose & Amylopectin)

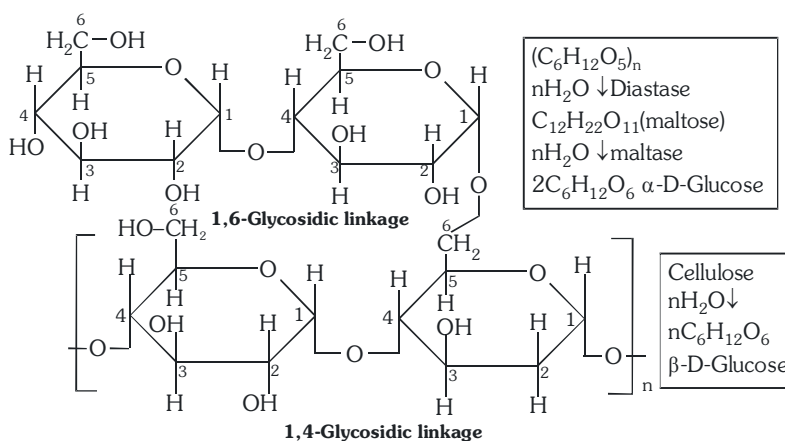
- **Amylose : (Straight Chain) :**



(α -1,4 Glycosidic linkage)

(i) Soluble in H_2O & gives blue colour with I_2

- **Amylopectin (Branch chain) : $(C_6H_{12}O_5)_n$**



REACTION OF GLUCOSE (OPEN CHAIN STRUCTURE)

* Fructose doesn't react with Br_2/H_2O

