

**Topic : General Organic Chemistry**
**Type of Questions**

Single choice Objective ('-1' negative marking) Q.1 to Q.5

Multiple choice objective ('-1' negative marking) Q.6 to Q.7

Subjective Questions ('-1' negative marking) Q.8

Match the Following (no negative marking) Q.9

(3 marks, 3 min.)

(4 marks, 4 min.)

(4 marks 5 min.)

(8 marks, 10 min.)

M.M., Min.

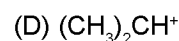
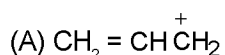
[15, 15]

[8, 8]

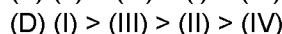
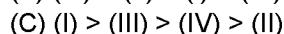
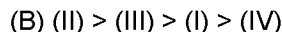
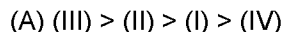
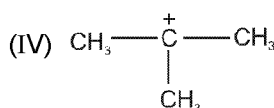
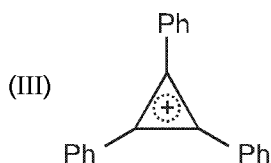
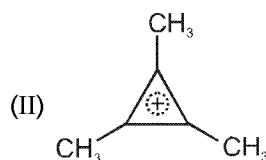
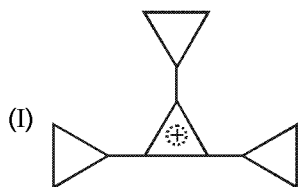
[4, 5]

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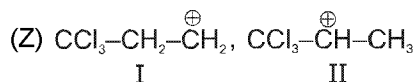
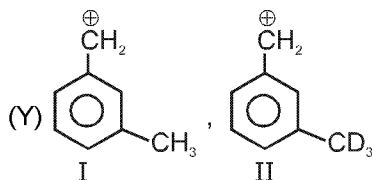
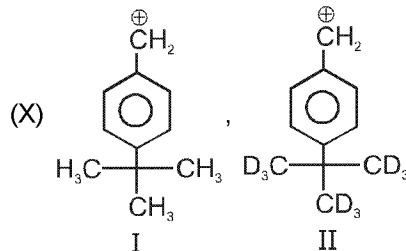
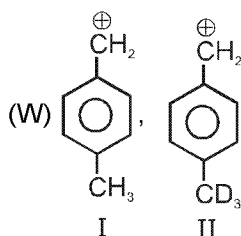
1. Which carbonium ion is highly stable ?



2. The correct order of stability of following carbocation is :



3. Observe each pair of cations. In which case (s) first is more stable than the second :



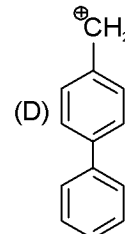
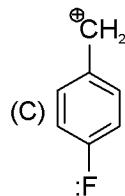
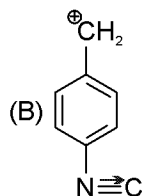
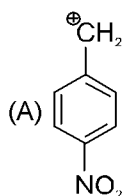
(A) Only in W

(B) Only in X and Y

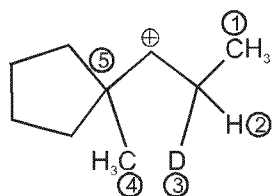
(C) Only in Z

(D) Only in W and Z

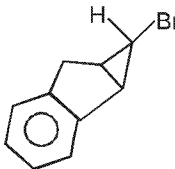
4. The most stable carbocation is :



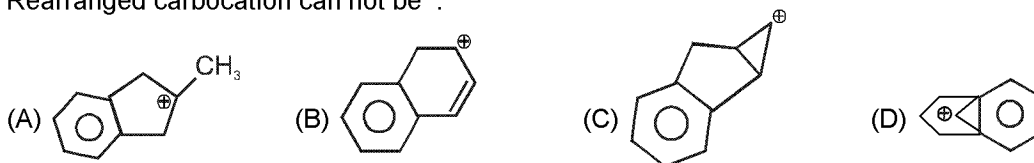
5. In the following carbocation; the most stable rearranged carbocation is formed by migration of group.



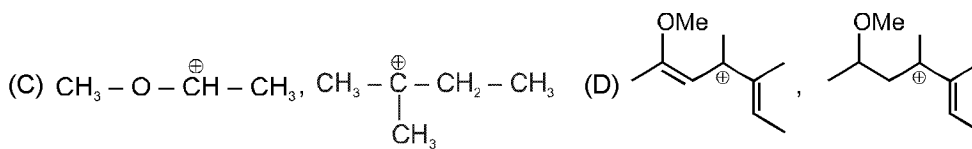
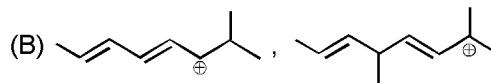
- (A)  $\text{CH}_3$  (1)      (B)  $\text{CH}_3$  (4)      (C) C—C bond (5)      (D) D (3)

6.   $\xrightarrow{\text{Ag}^+}$  Rearranged Carbocation + AgBr

Rearranged carbocation can not be :

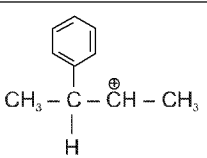
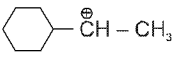
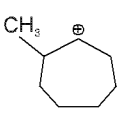
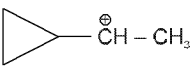
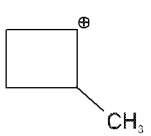
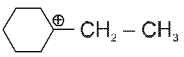
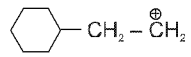
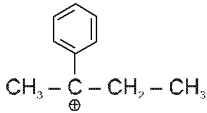


7. In which of the following first carbocation is more stable than second one ?



8. How many carbocations given below are more stable than sec. butyl carbocation  
t-butyl carbocation      Benzyl carbocation      Allyl carbocation  
Cyclopropenyl cation      Tropylium cation      n-butyl carbocation  
cyclopropylmethyl carbocation

9. Match the carbocation (I) with the most stable rearranged carbocation (II).

(I)	(II)
(A) 	(p) 
(B) 	(q) 
(C) 	(r) 
(D) 	(s) 

# Answer Key

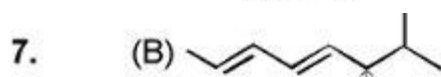
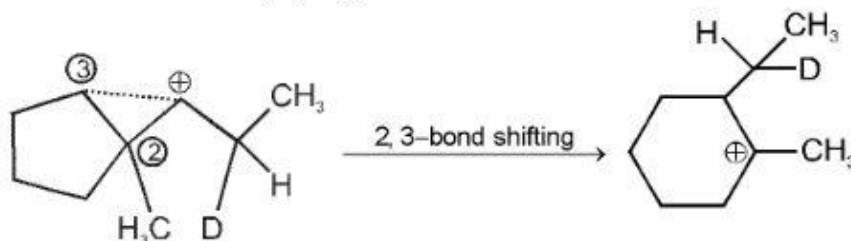
## DPP No. # 6

- |          |          |        |  |        |
|----------|----------|--------|--|--------|
| 1. (C)   | 2. (D)   | 3. (D) | 4. (D)                                   | 5. (C) |
| 6. (ACD) | 7. (BCD) | 8. 6   | 9. (A - s) ; (B - r) ; (C - q) ; (D - p) |        |

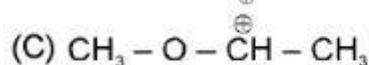
# Hints & Solutions

## DPP No. # 6

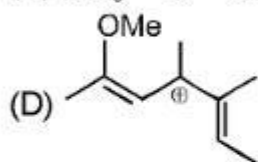
- (I) > (III) > (II) > (IV)
- In (W) since C-H bond is weaker than C-D bond so hyperconjugation stability is more in I.  
In (X) only +I effect is present which is more for  $-\text{C}(\text{CD}_3)$ .  
In (Y) only +I which is more for  $-\text{CD}_3$   
In (Z) -I effect of  $-\text{CCl}_3$  group will make II cation highly unstable.
- $e^-$  with drawing group decreases stability
- Due to C-C bond (5) migration it would be converted into 6-membered ring.



has extended conjugation.



has +M effect of  $-\text{OCH}_3$ .



after delocalisation gets +M effect of  $-\text{OMe}$ .

