



OPTIONAL NUMERICAL VALUE QUESTIONS (NVQs)

PHYSICS PART - II

- **OP21.** If the radius of curvature of the path of two particles of same masses are in the ratio 1 : 2, then in order to have constant centripetal acceleration, their velocity, should be in the ratio of _____.
- **OP22.** A refrigerator is to maintain eatables kept inside at 7°C. The coefficient of performance of refrigerator if room temperature is 38°C is _____.
- **OP23.** A capillary tube of radius r is immersed vertically in a liquid such that liquid rises in it to height h (less than the length of the tube). Mass of liquid in the capillary tube is m. If radius of the capillary tube is increased by 50%, then mass of liquid that will rise in the tube, is _____ times.
- **OP24.** 1/2 mole of helium is contained in a container at STP. How much heat energy is needed to double the pressure of the gas, (volume is constant). Heat capacity of gas is $3 \text{ jg}^{-1} \text{ K}^{-1}$.
- **OP25.** Velocity of sound waves in air is 330 m/s. For a particular sound wave in air, a path difference of 40 cm is equivalent to phase difference of 1.6π . The frequency of this wave Hz.

CHEMISTRY PART - II

- **OC46.** HCl is added to MnO_2 , PbO₂ and BaO. How many would not give H_2O_2 ?
- **OC47.** The γ -form of iron has *fcc* structure (edge length 386 pm) and β -form has *bcc* structure (edge length 290 pm). The ratio of density of γ -form and β -from is _____.
- **OC48.** Dry air was passed successively through a solution of 5 g of a solute in 80 g of water and then through pure water. The loss in weight of solution was 2.50 g and that of pure solvent 0.04 g. What is the molecular weight of the solute

- **OC49.** The frequency of radiation emitted when the electron falls from n=4 to n=1 in a hydrogen atom is $x \times 10^{15} \text{ s}^{-1}$ will be. (Given ionization energy of $H = 2.18 \times 10^{-18} J$ atom⁻¹ and $h = 6.625 \times 10^{-34} Js$)
- **OC50.** If 0.50 mol of $CaCl_2$ is mixed with 0.20 mol of Na_3PO_4 , the maximum number of moles of $Ca_3(PO_4)_2$ which can be formed, is _____.

MATHEMATICS PART - II

OM71. A biased coin with probability p, 0 , of heads to tossed until a head appears for the first time. If the probability that the number of tosses required is even 2

is
$$\frac{2}{5}$$
, then $p =$ _____.

OM72. The plane denoted by $P_1: 4x + 7x + 4z + 81 = 0$ is rotated through a right angle about its line of intersection with the plane $P_2: 5x + 3y + 10z = 25$. If the plane in its new position be denoted by P, and the distance of this plane from the origin is d, then the value of $\lfloor d/2 \rfloor$ (where $\lfloor k \rfloor$ represents greatest integer less then or equal to k) is _____.

OM73. The value of
$$\int_{0}^{\infty} |x-5| dx$$
 is _____.

OM74.
$$\lim_{x \to \infty} \left(\sqrt{x^2 + 8x + 3} - \sqrt{x^2 + 4x + 3} \right) = \underline{\qquad}$$

OM75. A batsman scores runs in 10 innings 38, 70, 48, 34, 42, 55, 63, 46, 54, 44, then the mean deviation about median is

C-15

SOLUTIONS

MOCK TEST 15

PHYSICS PART - II

OP21. (0.70) The centripetal acceleration, $a = \frac{v^2}{r} \Rightarrow r = \frac{v^2}{a}$ $\therefore r \propto v^2$ or $v \propto \sqrt{r}$ (if *a* is constant), $\Rightarrow \frac{v_1}{v_2} = \sqrt{\frac{r_1}{r_2}} = \sqrt{\frac{1}{2}}$ **OP22.** (9.03) Here, $T_1 = 38^{\circ}C = 38 + 273 = 311 \text{ K}$ $T_2 = 7^{\circ}C = 7 + 273 = 280 \text{ K}$ \therefore coefficient of performance of the refrigerator, $= \frac{T_2}{r_2} = \frac{280}{r_2} = 9.03$

$$T_1 - T_2 = 311 - 280$$

C-32

OP23. (1.5)
$$h = \frac{2T\cos\theta}{r\rho g} \Rightarrow h \propto \frac{1}{r} \Rightarrow \frac{h_2}{h_1} = \frac{r_1}{r_2} = \frac{2}{3}$$

 $\left(\because r_1 = r, r_2 = r + 50\% \text{ of } r = \frac{3}{2}r\right)$
New mass $m_2 = \pi r_2^2 h_2 \rho = \pi \left(\frac{3}{2}r_1\right)^2 \left(\frac{2}{3}h_1\right) \rho$
 $= \frac{3}{2} \left(\pi r_1^2 h_1\right) \rho = \frac{3}{2}m$
OP24. (1638) Here, $n = \frac{1}{2}, c_V = 3 \text{ J g}^{-1} \text{ K}^{-1}, M = 4 \text{g mol}^{-1}$
 $\therefore C_V = Mc_V = 4 \times 3 = 12 \text{ J mol}^{-1} \text{ K}^{-1}$
At constant volume $P \propto T$.
 $\therefore \frac{P_2}{P_1} = \frac{T_2}{T_1} = 2, T_2 = 2T_1$
Rise in temperature $\Delta T = T_2 - T_1 = 2T_1 - T_1 = T_1 = 273 \text{ K}$
Heat required, $\Delta Q = nC_V \Delta T = \frac{1}{2} \times 12 \times 273 = 1638 \text{ J}$
OP25. (660) From $\Delta x = \frac{\lambda}{2\pi} \Delta \phi$,
 $\lambda = 2\pi \frac{\Delta x}{\Delta \phi} = \frac{2\pi(0.4)}{1.6\pi} = 0.5 \text{ m}$

$$\Delta \phi = \frac{1.6\pi}{1.6\pi}$$
 f = $\frac{v}{\lambda} = \frac{330}{0.5} = 660$ Hz

CHEMISTRY PART - II

- **OC46.** (3) MnO_2 , PbO_2 and BaO will not give H_2O_2 with HCl. MnO_2 and PbO_2 will give Cl_2 and BaO will react with HCl to give BaCl₂ and water.
- **OC47.** (0.8481) γ -form fcc Z = 4 a = 386 pm β -form bcc Z = 2 a = 290 pm

$$\frac{\rho_{\gamma}}{\rho_{\beta}} = \frac{4M / N_{A} (386)^{3}}{2M / N_{A} (290)^{3}} = \frac{2(290)^{3}}{(386)^{3}} = 0.8481$$

OC48. (70.31)

OC49. (3.08)
$$E_{\text{ionisation}} = E_{\infty} - E_n = \frac{13.6Z_{eff}^2}{n^2} eV$$

 $= \left[\frac{13.6Z^2}{n_2^2} - \frac{13.6Z^2}{n_1^2}\right]$
 $E = hv = \frac{13.6 \times 1^2}{(1)^2} - \frac{13.6 \times 1^2}{(4)^2}; hv = 13.6 - 0.85$
 $\therefore h = 6.625 \times 10^{-34}$
 $v = \frac{13.6 - 0.85}{6.625 \times 10^{-34}} \times 1.6 \times 10^{-19} = 3.08 \times 10^{15} \text{ s}^{-1}.$

Mock Test

OC50. (0.1)
$$3\text{CaCl}_2 + 2\text{Na}_3\text{PO}_4 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 6\text{NaCl}$$

$$\therefore 2 \text{ Moles of } Na_3PO_4 = 3 \text{ mole of } CaCl_2$$
$$= 1 \text{ mole } Ca_3(PO_4)_2$$

 $\therefore \quad 0.2 \text{ mole of Na}_3 \text{PO}_4 = 0.3 \text{ mole of CaCl}_2$ $= 0.1 \text{ mole of Ca}_3(\text{PO}_4)_2$

MATHEMATICS PART - II

OM71. (0.33) Let X denotes the number of tosses required. Then $P(X = r) = (1 - p)^{r - 1}$. p, for r = 1, 2, 3..... Let *E* denote the event that the number of tosses required is even. Then $P(E) = P[(X = 2) \cup (X = 4) \cup (X = 6) \cup \dots]$ $P(E) = P(X = 2) + P(X = 4) + P(X = 6) + \dots$ $P(E) = (1-p)p + (1-p)^{3}p + (1-p)^{5}p + \dots = \frac{1-p}{2-p}$ But we are given that $P(E) = \frac{2}{5}$, then we get $p = \frac{1}{3}$. **OM72.** (7) 4x + 7y + 4z + 81 = 0...(i) 5x + 3y + 10z = 25...(ii) Equation of plane passing through their line of intersection is $(4x + 7y + 4z + 81) + \lambda(5x + 3y + 10z - 25) = 0$ or $(4+5\lambda)x + (7+3\lambda)y + (4+10\lambda)z + (81-25\lambda) = 0$...(iii) plane (iii) \perp to (i), so $4(4+5\lambda) + 7(7+3\lambda) + 4(4+10\lambda) = 0 \implies \lambda = -1$ From (iii), equation of plane is -x + 4y - 6z + 106 = 0...(iv) Distance of (iv) from (0, 0, 0)106 106 6

$$= \frac{1}{\sqrt{1+16+36}} = \frac{1}{\sqrt{53}} \approx 14.$$
$$\therefore \left[\frac{14.6}{2}\right] = [7.3] = 7.$$

OM73. (17)

OM74. (2)
$$\lim_{x \to \infty} \frac{4x}{\left(\sqrt{x^2 + 8x + 3} + \sqrt{x^2 + 4x + 3}\right)}$$
 (On rationalization)
$$= \lim_{x \to \infty} \frac{4}{\left(\sqrt{1 + \frac{8}{x} + \frac{3}{x^2}} + \sqrt{1 + \frac{4}{x} + \frac{3}{x^2}}\right)} = 2$$

OM75. (8.6) Arrange the given data in ascending order, We have 34, 38, 42, 44, 46, 48, 54, 55, 63, 70

Here, median
$$= M = \frac{46 + 48}{2} = 47$$

(:: $n = 10$, median is the mean of 5th and 6th term)
 \therefore Mean deviation $= \frac{\Sigma |x_i - M|}{n} = \frac{\Sigma |x_i - 47|}{10}$
 $= \frac{13 + 9 + 5 + 3 + 1 + 1 + 7 + 8 + 16 + 23}{10} = 8.6.$