

**NURTURE COURSE**

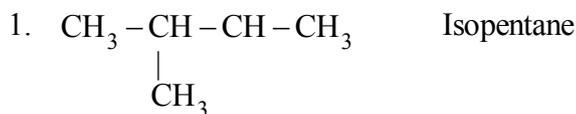
**NOMENCLATURE OF ORGANIC COMPOUND  
AND COMMON NAMES**



**NOMENCLATURE OF ORGANIC COMPOUND AND COMMON NAMES**

**-ESSENTIAL COMMON NAMES-**

**ALKANE**



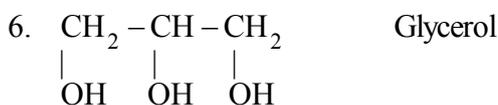
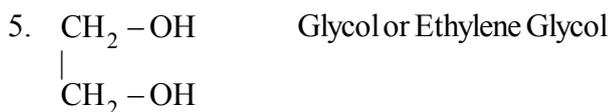
**ALKENE**



**ALKYL HALIDE**



**ALCOHOL**



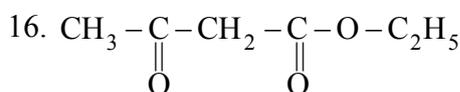
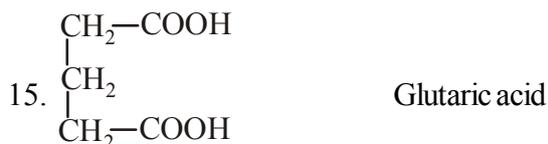
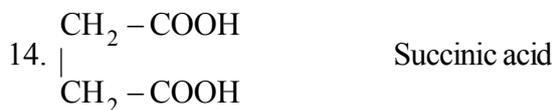
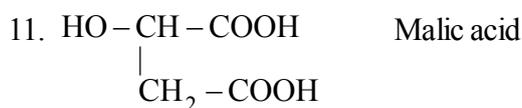
**ETHER**



**KETONE**

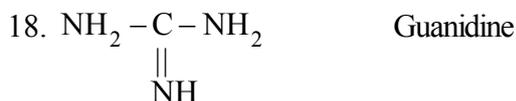


**CARBOXYLIC ACID**

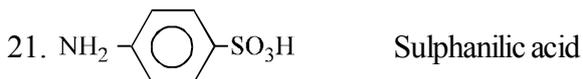
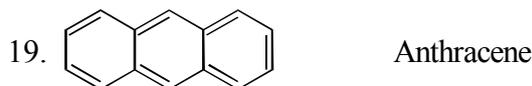


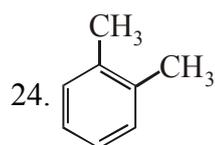
Aceto Acetic Ester (AAE) or Ethyl Aceto Acetate

**N-DERIVATIVES**

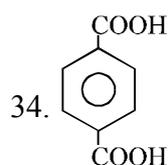


**AROMATIC COMPOUNDS**

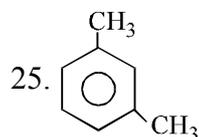




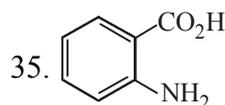
o-xylene



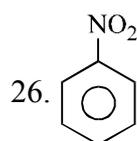
Terephthalic acid



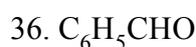
m-xylene



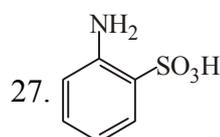
Anthranilic acid (o-aminobenzoic acid)



Nitrobenzene (oil of mirbane)

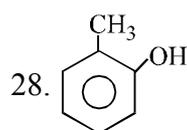


Benzaldehyde

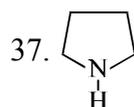


Ortho-anilic Acid

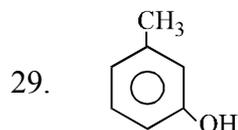
### HETEROCYCLIC COMPOUNDS



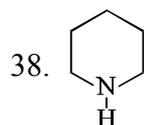
o-Cresol



Pyrrolidine



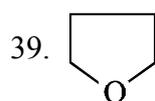
m-Cresol



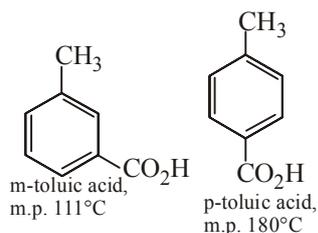
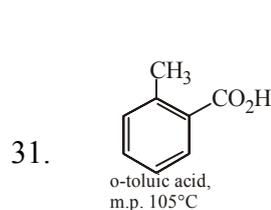
Piperidine



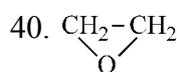
Perbenzoic acid



Tetrahydrofuran (THF)

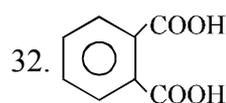


Toluic acids

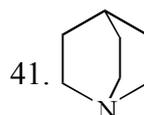


Oxirane or Ethylene Oxide or

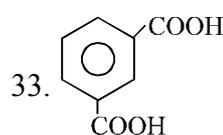
Oxo Cyclo Propane



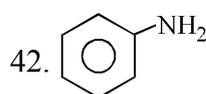
Phthalic acid



Quinuclidine



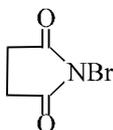
Isophthalic acid



Aniline

**SOME REAGENTS**

43. Grignard's reagent  $\text{RMgX}$
44. NBS  $\text{N-Bromosuccinimide}$

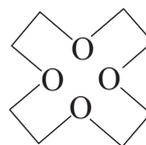


**POLAR PROTIC SOLVENTS**

45.  $\text{H-O-H}$  Water
46.  $\text{R-O-H}$  Alcohol
47. Phenol
48.  $\text{CH}_3\text{-C(=O)-OH}$  Acetic acid
49.  $\text{HF}$  Hydrogen Fluoride
50.  $\text{NH}_3$  Ammonia

**POLAR APROTIC SOLVENTS**

51. DMS Dimethyl sulphide  $\text{CH}_3\text{-S-CH}_3$
52. DMSO Dimethyl sulphoxide  $\text{Me}_2\text{S=O}$
53. HMPT Hexamethylphosphoramide  
or  
HMPTA  $\text{O=P-(NMe}_2)_3$
54. DMF Dimethyl formamide  
 $\text{H-C(=O)-NMe}_2$
55. Crown ethers Cyclic polyethers



(12 - C - 4)

**-DESIRABLE COMMON NAMES-****ALKANES**

1.  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CH} - \text{CH}_3 \\ | \quad | \\ \text{CH}_3 \quad \text{CH}_3 \end{array}$  Triptane
2.  $\begin{array}{c} -\text{CH}_2 - \text{CH}_2 - \text{CH} - \text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$  Isopentyl Group

**ALKENES**

3.  $\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH}_2$   $\alpha$ -Butylene
4.  $\text{CH}_3 - \text{CH} = \text{CH} - \text{CH}_3$   $\beta$ -Butylene
5.  $\begin{array}{c} \text{CH}_3 - \text{C} = \text{CH}_2 \\ | \\ \text{CH}_3 \end{array}$  Iso Butylene

**ALKYNES**

6.  $\text{HC} \equiv \text{CH}$  Purified Acetylene or Norcelyne
7.  $\text{CH}_3 - \text{C} \equiv \text{CH}$  Allylene

**ETHER**

8.  $\text{CH}_3\text{CH}(\text{OCH}_3)_2$  Methylal

**ALDEHYDE**

9.  $\begin{array}{c} \text{CHO} \\ | \\ \text{COOH} \end{array}$  Glyoxalic acid
10.  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CHO} \\ | \\ \text{CH}_3 \\ \text{or} \\ (\text{CH}_3)_3\text{C} - \text{CHO} \end{array}$  Pivaldehyde
11.  $(\text{CH}_3)_2\text{CHCHO}$  Isobutyraldehyde
12.  $\begin{array}{c} \text{CH}_3 - \text{C} - \text{C} - \text{CH}_3 \\ || \quad || \\ \text{O} \quad \text{O} \end{array}$  Dimethyl Glyoxal

13.  $\begin{array}{c} \text{CH}_3 - \text{C} - \text{C} - \text{H} \\ || \quad || \\ \text{O} \quad \text{O} \end{array}$  Methyl Glyoxal or Pyruvaldehyde

**KETONE**

14.  $\begin{array}{c} \text{CH}_3 \\ \diagdown \\ \text{C} = \text{CH} - \text{C} - \text{CH} = \text{C} \\ \diagup \quad || \quad \diagdown \\ \text{CH}_3 \quad \text{O} \quad \text{CH}_3 \end{array}$  Phorone
15.  $\begin{array}{c} \text{CH}_3 \\ \diagdown \\ \text{C} = \text{CH} - \text{C} - \text{CH}_3 \\ \diagup \quad || \\ \text{CH}_3 \quad \text{O} \end{array}$  Mesityl Oxide
16.  $\text{H}_2\text{C} = \text{C} = \text{O}$  Ketene

**CARBOXYLIC ACID**

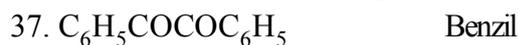
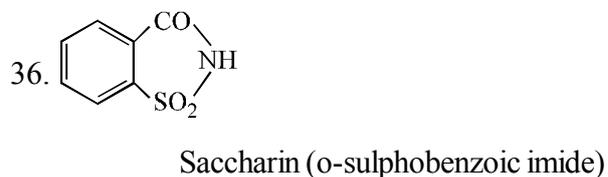
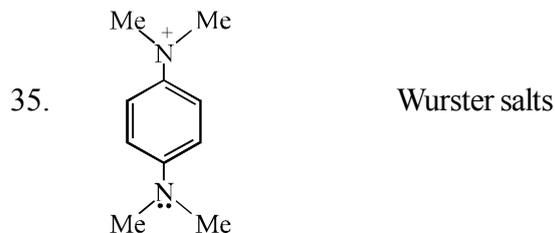
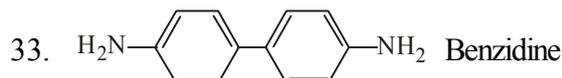
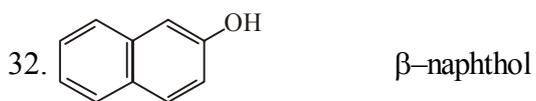
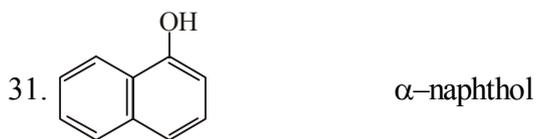
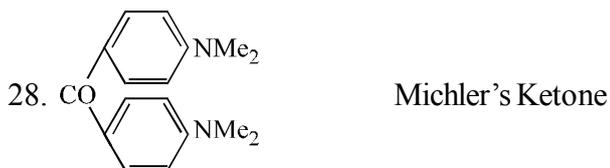
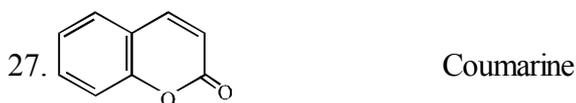
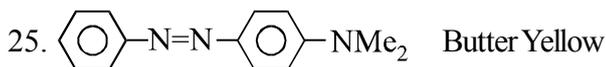
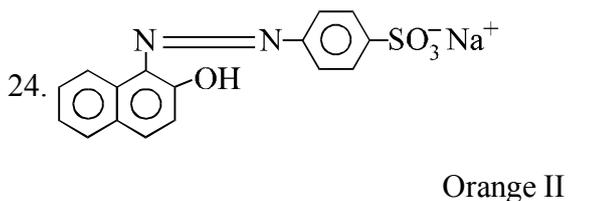
17.  $\text{CH}_3 - \text{CO} - \text{COOH}$  Pyruvic Acid
18.  $\begin{array}{c} \text{C}_6\text{H}_5 - \text{CH} - \text{COOH} \\ | \\ \text{OH} \end{array}$  Mandelic Acid
19.  $\text{NH}_2\text{COOH}$  Carbamic Acid (Amino formic Acid)

20.  $\begin{array}{c} \text{COOH} \\ | \\ \text{COOH} \end{array}$  Oxalic acid

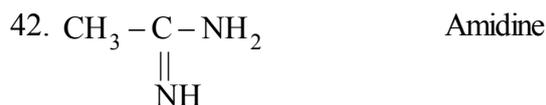
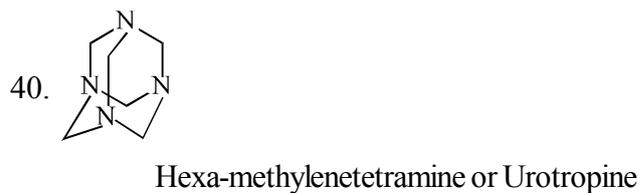
**ACID DERIVATIVES**

21.  $\begin{array}{c} \text{Cl} - \text{C} - \text{C} - \text{Cl} \\ || \quad || \\ \text{O} \quad \text{O} \end{array}$  Oxalyl Chloride
22.  $\text{NH}_2\text{COONH}_4$  Ammonium Carbamate
23.  $\begin{array}{c} \text{NH}_2 - \text{C} - \text{C} - \text{NH}_2 \\ || \quad || \\ \text{O} \quad \text{O} \end{array}$  Oxanamide

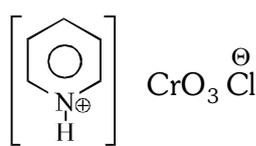
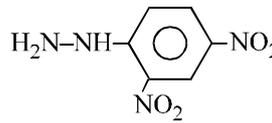
**AROMATIC COMPOUNDS**

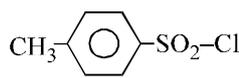


**HETEROCYCLIC COMPOUNDS**

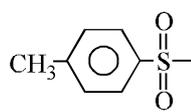
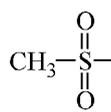
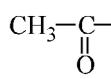
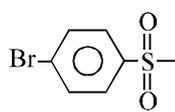
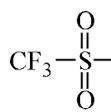


**SOME REAGENTS**

43. LAH	Lithium aluminium hydride : $\text{LiAlH}_4$
44. SBH	Sodium borohydride $\text{NaBH}_4$
45. PCC	Pyridinium chlorochromate 
46. Raney Nickel	Ni-Al alloy
47. Wilkinson's	Tris(Triphenylphosphine) catalyst chlororhodium (I) $(\text{PPh}_3)_3\text{RH}^+\text{Cl}^-$
48. Bayer's reagent	1% dil. alkaline aq.sol. of $\text{KMnO}_4$
49. Braddy's reagent 2,4 DNP	
50. Liemieux reagent	$\text{NaIO}_4 + \text{dil. alk. KMnO}_4$

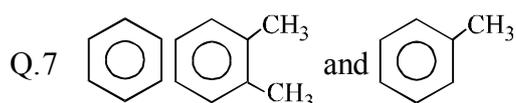
51. TEL	Tetra ethyl lead
52. Gillman's reagent	$\text{R}_2\text{CuLi}/[\text{R}_2\text{Cu}]^- \text{Li}^+$
53. Tollen's reagent	alk. sol. of $\text{AgNO}_3$
54. Fehling's reagent	alk. sol. of $\text{CuSO}_4$
55. Hinsberg's reagent	

**SOME GROUPS**

56. Ts	Tosyl	
57. Ms	Mesyl	
58. Ac	Acyl	
59. Bs	Brosyl	
60. Tf	Triflate	

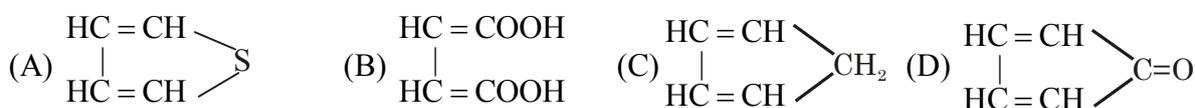
## EXERCISE # I

- Q.1 How many  $1^\circ$  carbon atom will be present in a simplest open chain hydrocarbon having two  $3^\circ$  and one  $2^\circ$  carbon atom ?  
 (A) 3 (B) 4 (C) 5 (D) 6
- Q.2 Alicyclic compounds are :  
 (A) Aromatic compounds (B) Aliphatic cyclic compounds  
 (C) Heterocyclic compounds (D) None of the above
- Q.3 How many  $1^\circ$ ,  $2^\circ$ ,  $3^\circ$  C atoms does 1, 3, 5-Trimethyl cyclohexane have?  
 (A) 3, 6, 0 (B) 3, 4, 2 (C) 0, 3, 6 (D) 3, 3, 3
- Q.4 The compound which has one isopropyl group is:  
 (A) 2,2,3,3-Tetramethyl pentane (B) 2,2-Dimethyl pentane  
 (C) 2,2,3-Trimethyl pentane (D) 2-Methyl pentane
- Q.5 Which of the following is the first member of ester homologous series?  
 (A) Ethyl ethanoate (B) Methyl ethanoate  
 (C) Methyl methanoate (D) Ethyl methanoate
- Q.6 A group closely related compounds which can be expressed by a general formula & in which two consecutive members differ by 14 in their molecular masses is called  
 (A) a heterogeneous series (B) a homologous series  
 (C) a homogeneous series (D) a electrochemical series



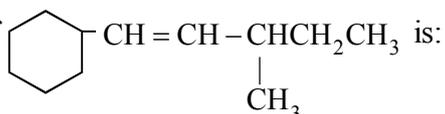
Number of secondary carbon atoms present in the above compounds are respectively:

- (A) 6,4,5 (B) 4,5,6 (C) 5,4,6 (D) 6,2,1
- Q.8 The molecular formula of the first member of the family of alkenynes and its name is given by the set  
 (A)  $C_3H_6$ , Alkene (B)  $C_5H_6$ , Pent-1-en-3-yne  
 (C)  $C_6H_8$ , Hex-1-en-5-yne (D)  $C_4H_4$ , Butenyne
- Q.9 Which of the following is a heterocyclic compound :

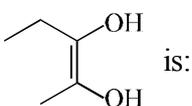


Q.10 The correct IUPAC name of the compound  $\text{CH}_3 - \text{CH}_2 - \overset{\text{CH}_3}{\text{C}} = \underset{\text{C}_2\text{H}_5}{\text{C}} - \text{CH} - \overset{\text{CH}_3}{\text{C}} - \text{CH}_2 - \text{CH}_2 - \text{CH}_3$  :

- (A) 5-Ethyl-3, 6-dimethyl non-3-ene (B) 5-Ethyl-4, 7-dimethyl non-3-ene  
(C) 4-Methyl-5, 7-diethyl oct-2-ene (D) 2,4-Ethyl-5-methyl oct-2-ene

Q.11 The IUPAC name of  is:

- (A) 1-Cyclohexyl-3-methyl pent-1-ene (B) 3-Methyl-5-cyclohexyl pent-1-ene  
(C) 1-Cyclohexyl-3-ethyl but-1-ene (D) 1-Cyclohexyl-3,4-dimethyl but-1-ene

Q.12 IUPAC name of  is:

- (A) But-2-ene-2,3-diol (B) Pent-2-ene-2,3-diol  
(C) 2-Methylbut-2-ene-2,3-diol (D) Pent-3-ene-3,4-diol

Q.13 IUPAC name of  $\text{CH}_2 = \text{CH} - \text{CN}$  is:

- (A) Ethenenitrile (B) Vinyl cyanide (C) Cyano ethene (D) Prop-2-enenitrile

Q.14 The IUPAC name of  $\text{CH}_3 \text{CH}_2 - \underset{\text{CH}_3}{\text{N}} - \text{CH}_2 \text{CH}_3$  is:

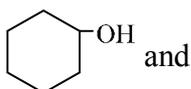
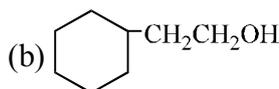
- (A) N-Methyl-N-ethyl ethanamine (B) Diethyl methanamine  
(C) N-Ethyl-N-methyl ethanamine (D) Methyl diethyl ethanamine

Q.15 The IUPAC name of acetyl acetone is :

- (A) Pentane-2,5- dione (B) Pentane -2,4-dione  
(C) Hexane-2,4-dione (D) Butane-2,4-dione

Q.16 When vinyl & allyl are joined each other, we get

- (A) Conjugated alkadiene (B) cumulative alkadiene  
(C) Isolated alkadiene (D) Allenes

Q.17 (a)  and (b) 

True statement for the above compounds is :

- (A) (a) is phenol while (b) is alcohol (B) Both (a) and (b) are primary alcohol  
(C) (a) is primary and (b) is secondary alcohol (D) (a) is secondary and (b) is primary alcohol

Q.18 The IUPAC name of the following structure  $(\text{CH}_3)\text{C}.\text{C}.\text{C}.\text{C}.\text{C}.\text{C}(\text{CH}_3)\text{CH}(\text{CH}_3)$  is:

- (A) 3-Methylhex-4-yn-2-ene (B) 3-Methylhex-2-en-4-yne  
(C) 4-Methylhex-4-en-4-yne (D) All are correct

Q.19 The IUPAC name of the following structure is  $[\text{CH}_3\text{CH}(\text{CH}_3)]_2 \text{C}(\text{CH}_2\text{CH}_3)\text{C}(\text{CH}_3) \text{C}(\text{CH}_2\text{CH}_3)_2$

- (A) 3,5-Diethyl-4,6-dimethyl-5-[1-methylethyl]hept-3-ene  
(B) 3,5-Diethyl-5-isopropyl-4,6-dimethylhept-2-ene  
(C) 3,5-Diethyl-5-propyl-4,6-dimethylhept-3-ene  
(D) None of these

Q.20 The correct IUPAC name of  $\text{CH}_3 - \text{CH}_2 - \underset{\text{CH}_2}{\underset{\parallel}{\text{C}}} - \text{COOH}$  is:

- (A) 2-Methyl butanoic acid (B) 2-Ethylprop-2-enoic acid  
(C) 2-Carboxybutene (D) None of the above

Q.21 The correct IUPAC name of 2-ethylpent-3-yne is:

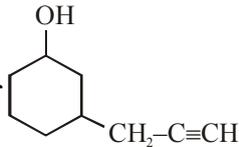
- (A) 3-Methyl hex-4-yne (B) 4-Ethyl pent-2-yne  
(C) 4-methyl hex-2 yne (D) None of these

Q.22 All the following IUPAC names are correct except:

- (A) 1-Chloro-1-ethoxy propane (B) 1-Amino-1-ethoxypropane  
(C) 1-Ethoxy-2-propanol (D) 1-Ethoxy-1-propanamine

Q.23 The IUPAC name of the compound  $\text{CH}_3\text{CH} = \text{CHCH} = \text{CHC} \equiv \text{CCH}_3$  is:

- (A) Octa-4,6-diene-2-yne (B) Octa-2,4-diene-6-yne  
(C) Oct-2-yne-4,6-diene (D) Oct-6-yne-2,4-diene

Q.24 The correct IUPAC name of 

- (A) 3-Cyclohexanol Propyne (B) 3-[3-Hydroxy Cyclohexyl] Propyne  
(C) 3-Propynyl Cyclohexanol (D) 3-(2-propynyl) Cyclohexanol

Q.25 The IUPAC name of  $\beta$ -ethoxy- $\alpha$ -hydroxy propionic acid (trivial name) is:

- (A) 1,2-Dihydroxy-1-oxo-3-ethoxy propane (B) 1-Carboxy-2-ethoxy ethanol  
(C) 3-Ethoxy-2-hydroxy propanoic acid (D) All above

Q.26 As per IUPAC rules, which one of the following groups, will be regarded as the principal functional group ?

- (A)  $-\text{C} \equiv \text{C}-$  (B)  $-\text{OH}$  (C)  $-\overset{\text{O}}{\parallel}{\text{C}}-$  (D)  $-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}$

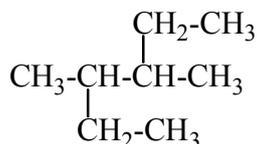


Q.33 The correct IUPAC name of compound  $\text{CH}_3 - \text{CH}_2 - \text{C} - \text{CH} - \text{CHO}$  is :



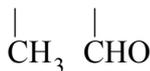
- (A) 2-Cyano-3-oxopentanal (B) 2-Formyl-3-oxopentanenitrile  
(C) 2-Cyanopentane-1,3-dione (D) 1,3-Dioxo-2-cyanopentane

Q.34 IUPAC name of compound



- (A) 2, 3-diethyl butane (B) 2-ethyl-3-methyl pentane  
(C) 3-methyl-2-ethyl pentane (D) 3,4-dimethyl hexane

Q.35 The IUPAC name of compound  $\text{CH}_3 - \overset{\text{O}}{\parallel} \text{C} - \text{CH} - \text{CH} - \overset{\text{CH}_3}{\text{CH}} - \text{CH}_3$  is:



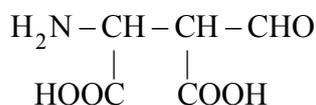
- (A) 3,5-Dimethyl-4-Formyl pentanone (B) 1-Isopropyl-2-methyl-4-oxo butanal  
(C) 2-Isopropyl-3-methyl-4-oxo pentanal (D) None of the above

Q.36 The IUPAC name of compound  $\text{CH}_3 - \overset{\text{HO}-\text{C}=\text{O}}{\text{C}} = \overset{\text{CH}_3}{\text{C}} - \text{H}$  is :



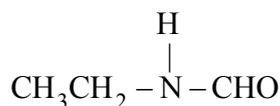
- (A) 2-Amino-3-chloro-2-methylpent-2-enoic acid (B) 3-Amino-4-chloro-2-methylpent-2-enoic acid  
(C) 4-Amino-3-chloro-2-methylpent-2-enoic acid (D) All of the above

Q.37 The IUPAC name of the structure is:



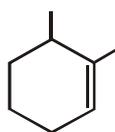
- (A) 3-Amino-2-formyl butane-1, 4-dioic acid (B) 3-Amino-2, 3-dicarboxy propanal  
(C) 2-Amino-3-formyl butane-1, 4-dioic acid (D) 1-Amino-2-formyl succinic acid

Q.38 One among the following is the correct IUPAC name of the compound



- (A) N-Formyl aminoethane (B) N-Ethyl formyl amine  
(C) N-Ethyl methanamide (D) Ethylamino methanal

Q.39 The IUPAC name of the structure is :



- (A) 1,2-Dimethyl-Cyclohexane (B) 1,6-Dimethyl-Cyclohexene  
(C) 1,2-Dimethyl-Cyclohex-2-ene (D) 2,3-Dimethyl-Cyclohexane

Q.40 The IUPAC name of  $C_6H_5CH=CH-COOH$  is :

- (A) Cinnamic acid (B) 1-Phenyl-2-carboxy ethane  
(C) 3-Phenyl prop-2-enoic acid (D) Dihydroxy-3-phenyl propionic acid

Q.41 The IUPAC name of  $BrCH_2-CH-CO-CH_2-CH_2CH_3$  is:



- (A) 2-Bromomethyl-3-oxohexanamide (B) 1-Bromo-2-amino-3-oxohexane  
(C) 1-Bromo-2-amino-n-propyl ketone (D) 3-Bromo-2-propyl propanamide

Q.42 IUPAC name will be  $\begin{array}{c} CH_2-CH-CH_2 \\ | \quad | \quad | \\ CN \quad CN \quad CN \end{array}$

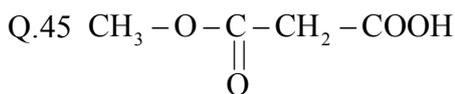
- (A) 1,2,3-Tricyano propane (B) Propane-1,2,3- trinitrile  
(C) 1,2,3-Cyano propane (D) Propane-1,2,3-tricarbonitrile

Q.43 The IUPAC name of compound is:

- (A) 3-Carbonyl methoxy -5- Ethanoyl oxy cyclohexan~~o~~ic acid  
(B) 3-Ethanoyl oxy -5- Methoxy carbonyl cyclohexane carboxylic acid  
(C) 5-Ethanoyl oxy -5- Methoxy carbonyl cyclohexanoic acid  
(D) 3-Methoxy carbonyl -5- Ethanoyl oxy cyclohexane carboxylic acid

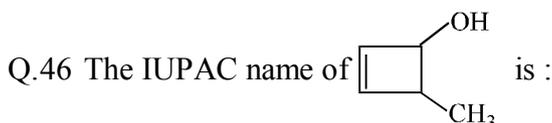
Q.44 The IUPAC name of  $CH_3-\overset{\overset{O}{||}}{C}-O-CH_2-\overset{\overset{O}{||}}{C}-OH$  is:

- (A) 1-Acetoxy acetic acid (B) 2-Acetoxy ethanoic acid  
(C) 2-Ethanoyloxyacetic acid (D) 2-Ethanoyloxyethanoic acid

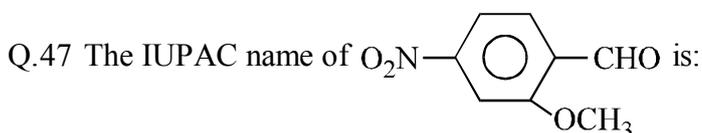


The correct IUPAC systematic name of the above compound is:

- (A) 2-Acetoxy ethanoic acid (B) 2-Methoxy carbonyl ethanoic acid  
(C) 3-Methoxy formyl ethanoic acid (D) 2-Methoxy formyl acetic acid

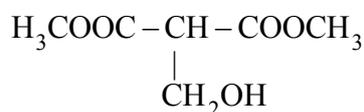


- (A) 3-Methyl cyclobut-1-ene-2-ol (B) 4-Methyl cyclobut-2-ene-1-ol  
(C) 4-Methyl cyclobut-1-ene-3-ol (D) 2-Methyl cyclobut-3-ene-1-ol

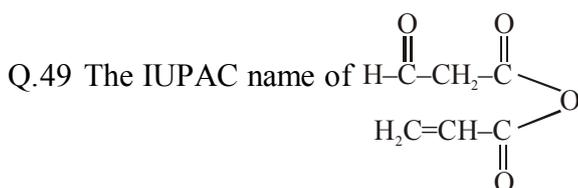


- (A) 2-Methoxy-4-nitro benzaldehyde (B) 4-Nitro anisaldehyde  
(C) 3-Methoxy-4-formyl nitro benzene (D) 2-Formyl-4-nitro anisole

Q.48 The IUPAC name of compound



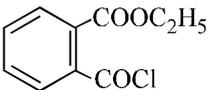
- (A) 2-(Hydroxy methyl) methyl propanedioate (B) Methyl-2-(hydroxy methyl) propanedioate  
(C) 2-(Hydroxy methyl) dimethyl propanedioate (D) None of these



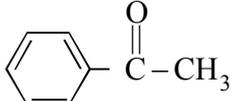
- (A) 2-Formyl ethanoic propanoic Anhydride (B) 2-Oxo-propanoic prop-2-enoic Anhydride  
(C) Prop-2-enoic-2-formyl propanoic Anhydride (D) 2-Formyl ethanoic prop-2-enoic Anhydride



- (A) 4,4-Di(formylmethyl) butanal (B) 2-(Formylmethyl) butane-1, 4-dicarbaldehyde  
(C) Hexane-3-acetal-1, 6-dial (D) 3-(Formylmethyl) hexane-1, 6-dial

Q.51 The IUPAC name of  is :

- (A) 2-Chlorocarbonyl ethylbenzoate      (B) 2-Carboxyethyl benzoyl chloride  
 (C) Ethyl-2-(chlorocarbonyl)benzoate      (D) Ethyl-1-(chlorocarbonyl)benzoate

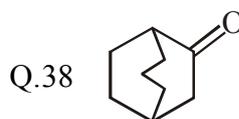
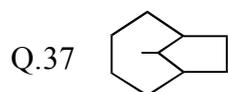
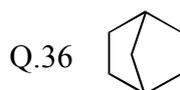
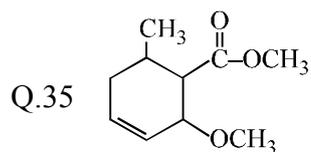
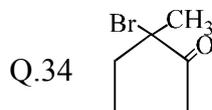
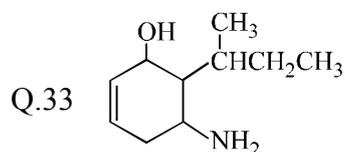
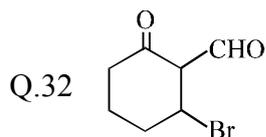
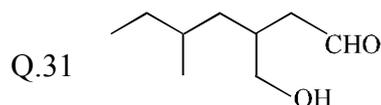
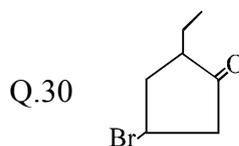
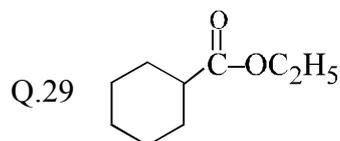
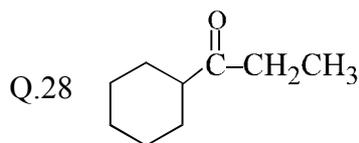
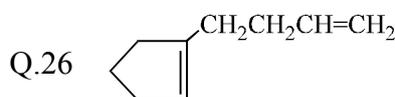
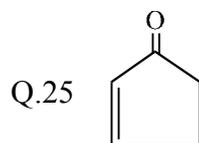
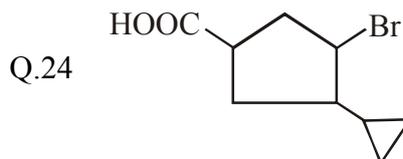
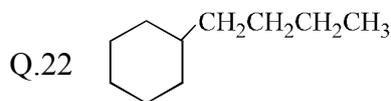
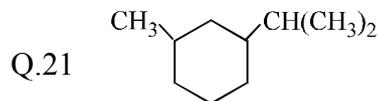
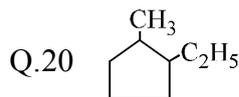
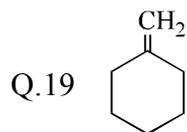
Q.52 The IUPAC name of  is:

- (A) Phenyl ethanone      (B) Methyl phenyl ketone  
 (C) Acetophenone      (D) Phenyl methyl ketone

Q.53 Structural formula of isopropyl methanoate is :

- (A)  $\text{CH}_3 - \underset{\text{O}}{\parallel}{\text{C}} - \text{O} - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_3$       (B)  $\text{H} - \underset{\text{O}}{\parallel}{\text{C}} - \text{O} - \text{CH}_2 - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_3$   
 (C)  $\text{CH}_3 - \underset{\text{O}}{\parallel}{\text{C}} - \text{O} - \text{CH}_2 - \underset{\text{CH}_3}{\text{CH}_2}$       (D)  $\text{H} - \underset{\text{O}}{\parallel}{\text{C}} - \text{O} - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_3$

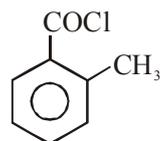




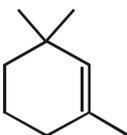
Q.39



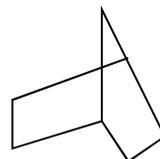
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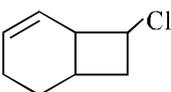
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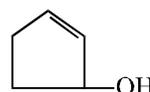
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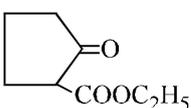
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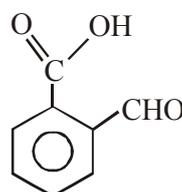
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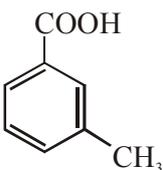
Q.45



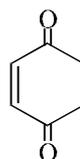
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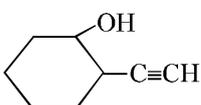
Q.47



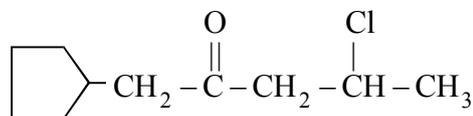
Q.48



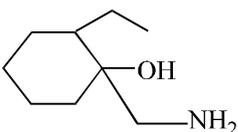
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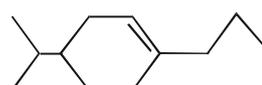
Q.50



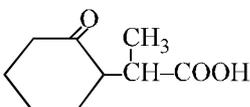
Q.51



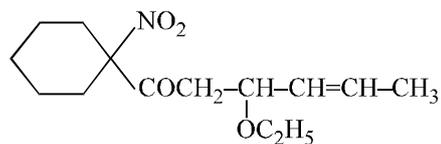
Q.52



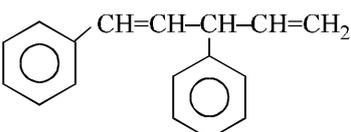
Q.53



Q.54



Q.55



## EXERCISE # III

- Q.1 Which of the following pairs have absence of carbocyclic ring in both compounds?  
 (A) Pyridine, Benzene (B) Benzene, Cyclohexane  
 (C) Cyclohexane, Furane (D) Furane, Pyridine
- Q.2 The commercial name of trichloroethene is:  
 (A) Westron (B) Perclene (C) Westrosol (D) Orlone
- Q.3 A substance containing an equal number of primary, secondary and tertiary carbon atoms is:  
 (A) Mesityl Oxide (B) Mesitylene (C) Maleic acid (D) Malonic acid
- Q.4 The IUPAC name of the compound Glycerine  $\text{CH}_2 - \text{CH} - \text{CH}_2$  is:  
 $\begin{array}{ccc} | & | & | \\ \text{OH} & \text{OH} & \text{OH} \end{array}$   
 (A) 1,2,3-Tri hydroxy propane (B) 3-Hydroxy pentane-1,5-diol  
 (C) 1,2,3-Hydroxy propane (D) Propane-1,2,3-triol
- Q.5 Which of the following is crotonic acid:  
 (A)  $\text{CH}_2 = \text{CH} - \text{COOH}$  (B)  $\text{C}_6\text{H}_5 - \text{CH} = \text{CH} - \text{COOH}$   
 (C)  $\text{CH}_3 - \text{CH} = \text{CH} - \text{COOH}$  (D)  $\begin{array}{c} \text{CH} - \text{COOH} \\ || \\ \text{CH} - \text{COOH} \end{array}$
- Q.6 The group of heterocyclic compounds is:  
 (A) Phenol, Furane (B) Furane, Thiophene  
 (C) Thiophene, Phenol (D) Furane, Aniline
- Q.7 **Column - I** (Common Name) **Column - II** (Structural formula)
- (A) Isooctane (P)  $\begin{array}{cc} \text{CH}_2 - \text{CH}_2 \\ | \quad | \\ \text{Cl} \quad \text{Cl} \end{array}$
- (B) Neopentane (Q)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CH} - \text{CH} - \text{CH}_3 \\ | \quad | \\ \text{CH}_3 \quad \text{CH}_3 \end{array}$
- (C) Ethylidene chloride (Geminal dihalide) (R)  $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \begin{array}{c} \text{CH} - \text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$
- (D) Ethylene Dichloride (Vicinal dihalide) (S)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$
- (T)  $\text{CH}_3 - \text{CH} \begin{array}{l} / \text{Cl} \\ \backslash \text{Cl} \end{array}$

**Q.8 Column - I**  
(Common Name)

- (A) Acetone
- (B) Acetaldehyde
- (C) Crotonaldehyde
- (D) Acrolein

**Column - II**  
(Structural formula)

- (P)  $\text{CH}_2 = \text{CH} - \overset{\text{O}}{\parallel} \text{C} - \text{H}$
- (Q)  $\text{CH}_3 - \overset{\text{O}}{\parallel} \text{C} - \text{CH}_3$
- (R)  $\text{CH}_3 - \text{CH} = \text{CH} - \overset{\text{O}}{\parallel} \text{C} - \text{H}$
- (S)  $\text{CH}_3 - \overset{\text{O}}{\parallel} \text{C} - \text{H}$

**Q.9 Column - I**  
(Common Name)

- (A)  $\begin{array}{c} \text{CHO} \\ | \\ \text{CHO} \end{array}$
- (B)  $\begin{array}{c} \text{CHO} \\ | \\ \text{H} - \text{C} - \text{OH} \\ | \\ \text{CH}_2 - \text{OH} \end{array}$
- (C)  $\text{H}_2\text{N} - \text{CH}_2 - \text{COOH}$
- (D)  $\begin{array}{c} \text{H} \\ | \\ \text{CH}_3 - \text{C} - \text{COOH} \\ | \\ \text{OH} \end{array}$

**Column - II**  
(Structural formula)

- (P) Lactic acid (In milk)
- (Q) Glyoxal
- (R) Glyceraldehyde
- (S) Glycine
- (T) Glycerol

**Q.10 Column - I**  
(Common Name)

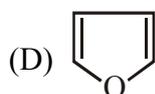
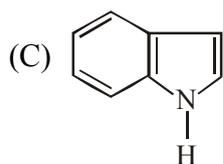
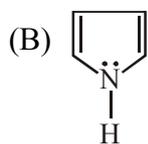
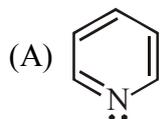
- (A) Fumaric acid
- (B) Adipic acid
- (C) Maleic acid
- (D) Tartaric acid

**Column - II**  
(Structural formula)

- (P)  $\begin{array}{c} \text{HO} - \text{CH} - \text{COOH} \\ | \\ \text{HO} - \text{CH} - \text{COOH} \end{array}$
- (Q)  $\begin{array}{c} \text{HC} - \text{COOH} \\ || \\ \text{HOOC} - \text{CH} \end{array}$
- (R)  $\begin{array}{c} \text{H} - \text{C} - \text{COOH} \\ || \\ \text{H} - \text{C} - \text{COOH} \end{array}$
- (S)  $\text{COOH}(\text{CH}_2)_4\text{COOH}$

Q.11 Column - I

(Common Name)



Column - II

(Structural formula)

(P) Pyrrole

(Q) Furan

(R) Thiophene

(S) Indol

(T) Pyridine

Q.12 Column - I

(Common Name)

(A) p-Cresol

(B) p-Xylene

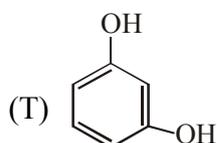
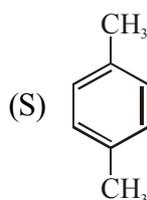
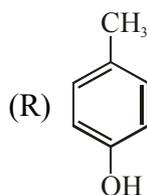
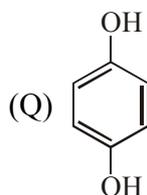
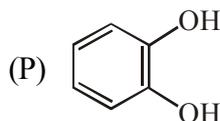
(C) Resorcinol

(D) Quinol

(E) Catechol

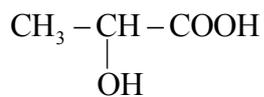
Column - II

(Structural formula)

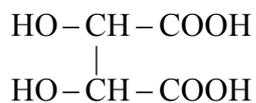


Q.13 Which of the following is not correctly matched:

(A) Lactic acid



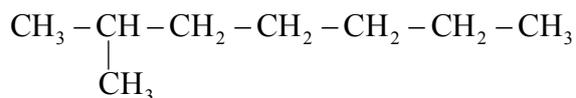
(B) Tartaric acid



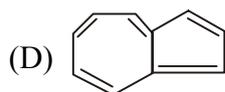
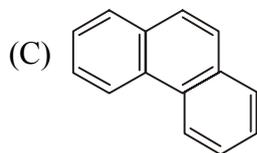
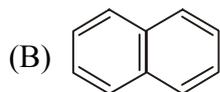
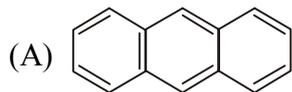
(C) Pivaldehyde



(D) Iso-octane



Q.14 **Column - I**



**Column - II**

(P) Phenanthrene

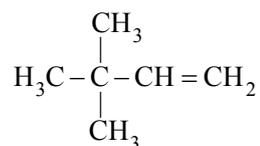
(Q) Anthracene

(R) Azulene

(S) Naphthalene

## EXERCISE # IV

Q.1 The IUPAC name of the compound having the formula is : [JEE 1984]



- (A) 3,3,3-trimethyl-1-propene  
 (B) 1,1,1-trimethyl-2-propene  
 (C) 3,3-dimethyl-1-butene  
 (D) 2,2-dimethyl-3-butene

Q.2 Write the IUPAC name of  $\text{CH}_3\text{CH}_2\text{CH}=\text{CH}\cdot\text{COOH}$  [JEE 1986]

Q.3 The IUPAC name of the compound  $\text{CH}_2=\text{CH}-\text{CH}(\text{CH}_3)_2$  is :

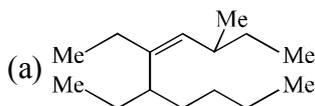
- (A) 1,1-dimethyl-2-propene (B) 3-methyl-1-butene  
 (C) 2-vinyl propane (D) None of the above

[JEE 1987]

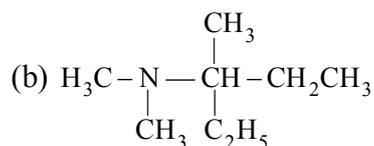
Q.4 The number of sigma and pi-bonds in 1-butene 3-yne are: [JEE 1989]

- (A) 5 sigma and 5 pi (B) 7 sigma and 3 pi  
 (C) 8 sigma and 2 pi (D) 6 sigma and 4 pi

Q.5 Write I.U.P.A.C name of following :



Me = methyl group



[JEE 1990]

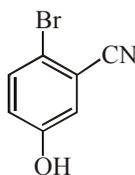
Q.6 Write IUPAC name of succinic acid. [JEE 1994]

Q.7 The IUPAC name of  $\text{C}_6\text{H}_5\text{COCl}$  is

- (A) Benzoyl chloride  
 (B) Benzene chloro ketone  
 (C) Benzene carbonyl chloride  
 (D) Chloro phenyl ketone

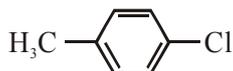
[JEE 2006]

Q.8 The IUPAC name of the following compound is [JEE 2009]



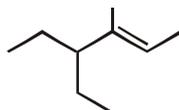
- (A) 4-Bromo-3-cyanophenol (B) 2-Bromo-5-hydroxybenzonitrile  
(C) 2-Cyano-4-hydroxybromobenzene (D) 6-Bromo-3-hydroxybenzonitrile

Q.9 The IUPAC name(s) of the following compound is(are) : [JEE 2017]



- (A) 4-methylchlorobenzene (B) 4-chlorotoluene  
(C) 1-chloro-4-methylbenzene (D) 1-methyl-4-chlorobenzene

10. The IUPAC name of the following compound is : [JEE Mains On\_line 2018]



- (A) 4-methyl-3-ethylhex-4-ene  
(B) 4,4-diethyl-3-methylbut-2-ene  
(C) 3-ethyl-4-methylhex-4-ene  
(D) 4-ethyl-3-methylhex-2-ene

**ANSWER KEY****EXERCISE # I**

Q.1	B	Q.2	B	Q.3	D	Q.4	D	Q.5	C	Q.6	B	Q.7	A
Q.8	D	Q.9	A	Q.10	A	Q.11	A	Q.12	B	Q.13	D	Q.14	C
Q.15	B	Q.16	C	Q.17	D	Q.18	B	Q.19	A	Q.20	B	Q.21	C
Q.22	B	Q.23	B	Q.24	D	Q.25	C	Q.26	D	Q.27	D	Q.28	C
Q.29	B	Q.30	B	Q.31	B	Q.32	B	Q.33	B	Q.34	D	Q.35	C
Q.36	B	Q.37	C	Q.38	C	Q.39	B	Q.40	C	Q.41	A	Q.42	D
Q.43	B	Q.44	D	Q.45	B	Q.46	B	Q.47	A	Q.48	B	Q.49	D
Q.50	D	Q.51	C	Q.52	A	Q.53	D						

**EXERCISE # II**

Q.1	4-Ethyl octane	Q.17	1,2-epoxy propane
Q.2	3-Ethyl-2,4-dimethyl pentane	Q.18	1,3,4-trimethyl cyclobutene
Q.3	5-Methyl hepta-1,3,6-triene	Q.19	Methylene cyclohexane
Q.4	Hepta-1,5-dien-3-yne	Q.20	1-ethyl-2-methylcyclopentane
Q.5	2-Isopropyl-4-methyl pent-1-ene or 4-Methyl-2-(methyl ethyl) pent-1-ene	Q.21	1-methyl-3-(methyl ethyl) cyclohexane or 1-isopropyl-3-methylcyclohexane
Q.6	3-Methoxypropene	Q.22	Butyl cyclohexane
Q.7	1-Hydroxybut-3-en-2-one	Q.23	Isopropylidenecyclopentane or 1-methyl ethylidene cyclopentane
Q.8	2-Ethylbut-2-en-1-ol	Q.24	3-Bromo-4-cyclopropyl cyclopentane carboxylic acid
Q.9	3-nitroprop-2-en-1-ol	Q.25	Cyclopent-2-en-1-one
Q.10	4-hydroxyhex-5-en-1-yn-3-one	Q.26	1-(3-butenyl) cyclopentene
Q.11	4,6-Bis-[1,1-Dimethyl ethyl] Nonane	Q.27	1,2-diethenyl cyclohexene
Q.12	2-Formyl pentane nitrile	Q.28	1-cyclohexyl-1-propanone
Q.13	2,2,6,7-tetramethylcatane	Q.29	Ethyl cyclohexanecarboxylate
Q.14	3-Ethyl-4,6-dimethyloctane	Q.30	4-Bromo-2-ethyl cyclopentanone
Q.15	5-Methyl cyclohexa-1,3-diene	Q.31	3-(hydroxymethyl)-5-methylheptanal
Q.16	4-Ethyl Pent-4-en-2-amine		

- Q.32** 2-Bromo-6-oxocyclohexanecarbaldehyde
- Q.33** 5-amino-6-(1-methyl propyl)  
cyclo hex-2-enol
- Q.34** 2-bromo-2-methyl cyclopentanone
- Q.35** Methyl-2-methoxy-6-methyl-3- cyclohexene  
carboxylate
- Q.36** Bicyclo(2,2,1)heptane
- Q.37** 9-methyl bicyclo(4,2,1) nonane
- Q.38** Bicyclo [3,2,2] Non-6-one
- Q.39** spiro(4,5) decane
- Q.40** 2-Methyl Benzoyl Chloride
- Q.41** 1,3,3-Trimethyl cyclohexene
- Q.42** Bicyclo(2,2,1) heptane
- Q.43** 8-chloro bicyclo(4,2,0) oct-2-ene
- Q.44** 2-cyclopenten-1-ol
- Q.45** Ethyl-2-oxo cyclo pentane carboxylate
- Q.46** 2-Formyl Benzoic acid
- Q.47** 3-Mthyl Benzoic acid
- Q.48** Cyclohex-2-en-1,4-dione
- Q.49** 2-ethynyl cyclohexanol
- Q.50** 4-chloro-1-cyclopentyl pentane-2-one
- Q.51** 1-Amino methyl-2-ethyl cyclohexanol
- Q.52** 4-isopropyl -1-propyl cyclohexene  
or 4-(methyl ethyl)-1-propyl cyclohexene
- Q.53** 2-(2-oxo-cyclohexyl) propanoic acid
- Q.54** 3-ethoxy-1(1-nitrocyclohexyl)-hex-4-en-1-one
- Q.55** 1,3-diphenyl-1,4-pentadiene

## EXERCISE # III

- Q.1 D    Q.2 C    Q.3 B    Q.4 D    Q.5 C    Q.6 B
- Q.7 (A)-Q ; (B)-S ; (C)-T ; (D)-P    Q.8 (A)-Q ; (B)-S ; (C)-R ; (D)-P
- Q.9 (A)-Q ; (B)-R ; (C)-S ; (D)-P    Q.10 (A)-Q ; (B)-S ; (C)-R ; (D)-P
- Q.11 (A)-T ; (B)-P ; (C)-S ; (D)-Q    Q.12 (A)-R ; (B)-S ; (C)-T ; (D)-Q ; (E)-P
- Q.13 D    Q.14 (A)-Q ; (B)-S ; (C)-P ; (D)-R

## EXERCISE # IV

- Q.1 C
- Q.2  $\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH} - \text{COOH}$   
           5      4      3      2      1  
 2-pentene-1-oic acid and or 2-pentenoic acid
- Q.3 B    Q.4 B
- Q.5 (a) 5,6-diethyl-3-methyl-dec-4-ene  
 (b) N,N, 3-trimethyl-3-pentanamine
- Q.6 Butane-1,4-dioic acid    Q.7 C    Q.8 B    Q.9 B,C
10. D







