

# Chapter 9

# POLYMERS



## CLASSIFICATION OF POLYMERS

1. "Based on source", polymers are classified as - (3)
2. Natural polymers ex - (5)
3. Semi-synthetic polymers ex - (2)
4. Cellulose acetate is also called -
5. "Based on structure", polymers classified as - (3)
6. Cross-linked polymer also called -
7. PVC is an ex. of linear/branched/network polymers.
8. Linear polymers ex - (2)
9. Branched-chain polymers ex - (1)
10. Cross-linked polymers ex - (2) (NEET)
11. \_\_\_\_\_ polymers are usually formed from bi-functional and tri-functional monomers. (NEET)
12. "Based on Mode of Polymerisation", polymers are classified as - (2)
13. Homopolymers ex - (1)
14. Copolymers ex - (2)
15. Monomers is Buna-S are - (NEET)
16. Monomers in Buna-N are -
17. Condensation polymers ex - (3)
18. Terylene is also called -
19. "Based on Molecular Forces", polymers are classified as - (4)
20. Elastomers are held together by weak/strong intermolecular forces.
21. Elastomers ex - (3)
22. Fibers possess high tensile strength and high modulus. T/F
23. Fibers have strong intermolecular forces. T/F
24. Ex. of intermolecular forces present in fibers are -
25. Fibers ex - (2)
26. Thermoplastic possess intermolecular forces of attraction intermediate between elastomers and fibres. T/F
27. Thermoplastic polymers ex - (3)
28. Thermosetting polymers are slightly branched. T/F
29. Thermosetting polymers can be reused. T/F
30. Thermosetting polymers ex - (2)
31. "Based on Growth Polymerisation", polymers are classified as - (2)
32. Chain growth polymers are actually condensation polymers. T/F

# TYPES OF POLYMERISATION REACTIONS

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## ADDITION POLYMERISATION

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33. Chain growth polymerisation is the most common mode of formation of polymers. T/F
34. Ex. of free radical generating initiator - (3)
35. LDP full form -
36. LDP is made under pressure of \_\_\_\_\_ to \_\_\_\_\_ atm at a temperature of \_\_\_\_\_- \_\_\_\_\_ K in the presence of traces of \_\_\_\_\_ or \_\_\_\_\_
37. LDP have very little branching. T/F
38. LDP is used to manufacture - (3)
39. Catalyst used to make HDP is -
40. Ziegler-Natta catalyst consist of - (2)
41. Pressure used to make HDP is less as compared to pressure used to make LDP. T/F
42. Pressure used to make HDP is \_\_\_\_\_ atm and temp is \_\_\_\_\_ - \_\_\_\_\_ K
43. HDP have linear molecules. T/F
44. HDP is chemically inert and more tough and hard. T/F
45. HDP is used to manufacture - (3)
46. Monomer of teflon is - (NEET)
47. Tetrafluoroethene is heated with \_\_\_\_\_ or \_\_\_\_\_ catalyst to form teflon.
48. \_\_\_\_\_ is used for non-stick surface coated utensils. (NEET)
49. Acrylonitrile formula -
50. Polyacrylonitrile is used as a substitute for \_\_\_\_\_ (NEET)
51. Polyacrylonitrile is used to make fibers like \_\_\_\_\_ or \_\_\_\_\_

## CONDENSATION POLYMERISATION

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52. Polyamides ex - (2) (NEET)
53. Nylon 6,6 monomers - (2) (NEET)
54. Adipic acid formula -
55. Nylon 6,6 is used in making - (2)
56. Nylon 6 monomer - (NEET)
57. Nylon 6 is used in making - (3)
58. Polyesters ex - (1) (NEET)
59. Monomers used in terylene - (2) (NEET)
60. Dacron is obtained by heating the monomers at \_\_\_\_\_- \_\_\_\_\_ K in the presence of \_\_\_\_\_ catalyst.
61. Dacron fibre is crease resistant. T/F
62. It is used as glass reinforcing material in safety helmets. T/F

63. Novolac monomers are - (2)  
64. Novolac is used in \_\_\_\_\_  
65. Novolac on heating with formaldehyde undergo cross linking to form \_\_\_\_\_ (NEET)  
66. Electrical switches are made of -  
67. Unbreakable crockery is manufactured by -

## COPOLYMERISATION & RUBBER

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68. Buna-S is used to manufacture - (4)  
69. Rubber latex is colloidal dispersion of \_\_\_\_\_ in \_\_\_\_\_  
70. Natural rubber is linear polymer of \_\_\_\_\_ (NEET)  
71. Cis/trans isoprene is present in natural polymer. (NEET)  
72. Natural polymer is chemically called - (NEET)  
73. Cis-polyisoprene chains are held together by \_\_\_\_\_ interactions.  
74. Vulcanisation of rubber is done by heating rubber with \_\_\_\_ at temp between \_\_\_\_\_-\_\_\_\_\_ K..  
75. Rubber gets stiffened on vulcanisation. T/F (NEET)  
76. In manufacture of tyre rubber, \_\_\_\_% of sulphur is used.  
77. Synthetic rubber ex - (2)  
78. Neoprene monomer is - (NEET)  
79. \_\_\_\_\_ is used to manufacture conveyor belts, gaskets and hoses.  
80. \_\_\_\_\_ is used in making oil seals, tank lining.

## BIODEGRADABLE POLYMERS & COMMERCIALLY IMPORTANT POLYMERS

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81. Ex. of biodegradable polymers - (2) (NEET)  
82. PHBV full form -  
83. PHBV monomers - (2)  
84. 'Bute' means 4 and 'Valler' means -  
85. 'Capro' means -  
86. Nylon 2-nylon 6 monomers - (2)  
87. \_\_\_\_\_ is used in specialty packaging, orthopaedic devices and in controlled release of drugs.  
88. Glyptal monomers

## DIGAQS

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- DigaQ1. Melamine Structure  
DigaQ2. Chloroprene Structure
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## ANSWERS

### • CLASSIFICATION OF POLYMERS

1. Natural, semi-synthetic, synthetic
2. Proteins, cellulose, starch, some resins and rubber
3. cellulose acetate and cellulose nitrate
4. Rayon
5. Linear, Branched, Cross-linked
6. Network polymers
7. Linear
8. Polythene, PVC
9. LDP (low density polythene)
10. Bakelite, melamine
11. Cross linked
12. Addition, condensation
13. Polythene
14. Buna-S, Buna-N
15. 1,3-Butadiene + Styrene
16. 1,3-Butadiene + Acrylonitrile
17. Terylene, nylon 6, 6, nylon 6
18. Dacron
19. Elastomers, fibres, thermoplastic polymers, thermosetting polymers
20. Weak
21. Buna-S, Buna-N, neoprene
22. T
23. T
24. Hydrogen bonding
25. Terylene, nylon 6,6
26. T
27. Polythene, polystyrene, polyvinyls
28. F
29. F
30. Bakelite, urea-formaldehyde resins
31. Chain growth polymers and step growth polymers
32. F

### • ADDITION POLYMERISATION

33. T
34. benzoyl peroxide, acetyl peroxide, tert-butyl peroxide
35. Low density polythene
36. 1000 to 2000 atm, 350-570 K, dioxygen or a peroxide initiator
37. F
38. squeeze bottles, toys and flexible pipes
39. Ziegler-Natta catalyst
40.  $\text{TiCl}_4$ ,  $\text{Al}(\text{C}_2\text{H}_5)_3$
41. T
42. 6-7 atm, 333-343 K
43. T
44. T
45. buckets, dustbins, bottles, pipes
46. Tetrafluoroethene
47. Free radical or persulfate catalyst
48. Teflon
49.  $\text{CH}_2=\text{CHCN}$
50. Peroxide catalyst
51. Orlon or acrilan

### • CONDENSATION POLYMERISATION

52. Nylon 6 & Nylon 6,6
53. Hexamethylenediamine and adipic acid
54.  $\text{HOOC} - (\text{CH}_2)_4 - \text{COOH}$
55. Sheets, bristles for brushes
56. Caprolactam
57. Type cords, fabrics and ropes
58. Terylene
59. ethylene glycol and terephthalic acid

60. 420-460 K, zinc acetate-antimony trioxide catalyst

61. T

62. T

63. Phenol and formaldehyde

64. Paints

65. Bakelite

66. Bakelite

67. Melamine-formaldehyde polymer

### • COPOLYMERISATION & RUBBERS

68. autotyres, floor tiles, footwear components, cable insulation

69. Rubber in water

70. Isoprene

71. Cis

72. cis-1,4-polyisoprene

73. van der Waals

74. S, 373-415 K

75. T

76. 5%

77. Neoprene, Buna-S

78. Chloroprene

79. Neoprene

80. Buna-N

### • BIODEGRADABLE POLYMERS & COMMERCIALY IMPORTANT POLYMERS

81. PHBV, Nylon 2-nylon 6

82. Poly  $\beta$ -hydroxybutyrate - co- $\beta$ -hydroxy valerate

83. 3-Hydroxybutanoic acid + 3-Hydroxypentanoic acid

84. 5

85. 6

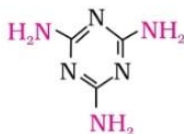
86. Glycine + amino caproic acid

87. PHBV

88. Ethylene glycol + Phthalic acid

### • DIGAQS

DigaQ 1.



DigaQ 2.

