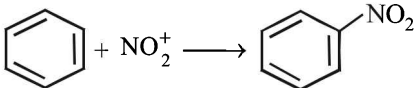


Aldehydes, Ketones and Carboxylic Acids



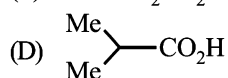
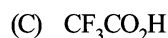
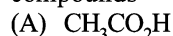
Conceptual MCQs

- Which of the following is disproportionation reaction?
 - $2\text{HCHO} \xrightarrow{\text{NaOH}} \text{CH}_3\text{OH} + \text{HCOONa}$
 - $2\text{CH}_3\text{CHO} \xrightarrow{\text{NaOH}} \text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{CHO}$
 - 
 $\text{C}_6\text{H}_6 + \text{NO}_2^+ \longrightarrow \text{C}_6\text{H}_5\text{NO}_2$
 - Both (a) & (b)
- Ketones $[\text{R}-\text{C}(=\text{O})-\text{R}_1]$, where $\text{R} = \text{R}_1 = \text{alkyl groups}$ can be obtained in one step by :
 - oxidation of primary alcohols.
 - hydrolysis of esters.
 - oxidation of secondary alcohols.
 - reaction of acid halides with alcohols.
- Hydride ion transfer takes place in :
 - Frankland method
 - Wurtz reaction
 - Cannizzaro's reaction
 - Wolf-Kishner reduction
- Calcium formate on distillation gives :
 - HCOOH
 - CH_3COOH
 - CH_3CHO
 - HCHO
- The cyanohydrin of a compound on hydrolysis gives an optically active α -hydroxy acid. The compound is :
 - diethyl ketone
 - formaldehyde
 - acetaldehyde
 - acetone
- Formalin is an aqueous solution of :
 - fluorescein
 - formic acid
 - formaldehyde
 - furfuraldehyde
- The catalyst used in Rosenmund's reduction is :
 - HgSO_4
 - Pd/BaSO_4
 - anhydrous AlCl_3
 - anhydrous ZnCl_2
- Acetaldehyde reacts with :
 - electrophiles only.
 - nucleophiles only.
 - free radicals only.
 - both electrophiles and nucleophiles.
- Strongest acid among the following is :
 - CF_3COOH
 - CBr_3COOH
 - CH_3COOH
 - CCl_3COOH
- Pinacolone is :
 - 2, 3-dimethyl-2 3-butanediol
 - 3, 3-dimethyl-2 butanone
 - 1-phenyl-2propanone
 - 1, 1-diphenyl-2-ethandiol.
- Which one of the following compounds will not react with CH_3MgBr ?
 - Ethyl acetate
 - Acetone
 - Dimethyl ether
 - Ethanol
- Appropriate reducing agent for the following conversion is—

$$\text{CH}_2 = \text{CH} - \text{CH}_2 - \overset{\text{O}}{\parallel}{\text{C}} - \text{H} \longrightarrow \text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2\text{OH}$$
 - $\text{LiAlH}_4/\text{H}_2\text{O}$
 - $\text{NaBH}_4/\text{H}_2\text{O}$
 - $\text{Na} + \text{C}_2\text{H}_5\text{OH}$
 - $\text{B}_2\text{H}_6/\text{H}^+$
- In Cannizzaro reaction given below

$$2\text{PhCHO} \xrightarrow{:\text{OH}^-} \text{PhCH}_2\text{OH} + \text{PhC}\ddot{\text{O}}^-$$
 the slowest step is :
 - the transfer of hydride to the carbonyl group
 - the abstraction of proton from the carboxylic group
 - the deprotonation of PhCH_2OH
 - the attack of $:\text{OH}^-$ at the carboxyl group
- Aldol condensation would not occur in :
 - CH_3COCH_3
 - $\text{CH}_3\text{CH}_2\text{CHO}$
 - HCHO
 - CH_3CHO
- Iodoform can be prepared from all except :
 - Ethyl methyl ketone
 - Isopropyl alcohol
 - 3-Methyl 2-butanone
 - Isobutyl alcohol
- Cannizzaro reaction occurs with :
 - $\text{CH}_3 - \text{CH}_2\text{OH}$
 - $\text{C}_6\text{H}_5\text{CHO}$
 - CH_3CHO
 - $\text{CH}_3 - \text{CO} - \text{CH}_3$
- In which of the following, the number of carbon atoms does not remain same when carboxylic acid is obtained by oxidation?
 - CH_3COCH_3
 - $\text{CCl}_3\text{CH}_2\text{CHO}$
 - $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$
 - $\text{CH}_3\text{CH}_2\text{CHO}$
- Which of the following compound will show positive silver mirror test ?
 - HCOOH
 - $\text{CH}_3(\text{CHOH})_3\text{CHO}$
 - $\text{CH}_3\text{CO}(\text{CHOH})\text{CH}_3$
 - Both (a) and (b)

19. The correct order of increasing acid strength of the compounds



is :

- (a) $\text{D} < \text{A} < \text{B} < \text{C}$ (b) $\text{A} < \text{D} < \text{B} < \text{C}$
 (c) $\text{B} < \text{D} < \text{A} < \text{C}$ (d) $\text{D} < \text{A} < \text{C} < \text{B}$
20. 2-pentanone and 3-pentanone can be distinguished by :
- (a) Cannizzaro's reaction (b) Aldol condensation
 (c) Iodoform reaction (d) Clemmensen's reduction

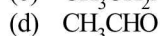
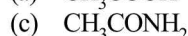


Application Based MCQs

21. Which of the following statement is incorrect about the reaction of ammonia derivatives with carbonyl compounds?

- (a) pH of solution is maintained between 4 to 5.
 (b) Addition of ammonia derivatives occur followed by elimination of H_2O .
 (c) At very low pH (less than 3) ammonia derivatives are protonated and do not act as nucleophile.
 (d) At very high pH reaction becomes explosive.

22. Identify the product C in the series.



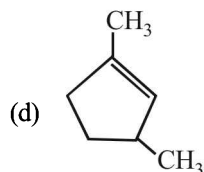
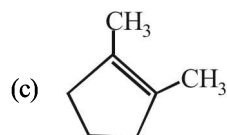
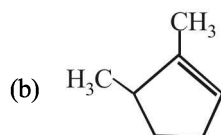
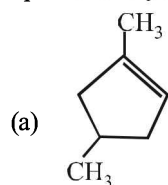
23. Ozonolysis of an organic compound 'A' produces acetone and propionaldehyde in equimolar mixture. Identify 'A' from the following compounds:

- (a) 1 - Pentene
 (b) 2 - Pentene
 (c) 2 - Methyl - 2 - pentene
 (d) 2 - Methyl - 1 - pentene

24. Benzaldehyde reacts with ethanoic KCN to give :

- (a) $\text{C}_6\text{H}_5\text{CHOHCN}$ (b) $\text{C}_6\text{H}_5\text{CHOHCOC}_6\text{H}_5$
 (c) $\text{C}_6\text{H}_5\text{CHOHCOOH}$ (d) $\text{C}_6\text{H}_5\text{CHOHCHOHC}_6\text{H}_5$

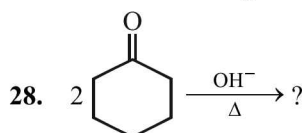
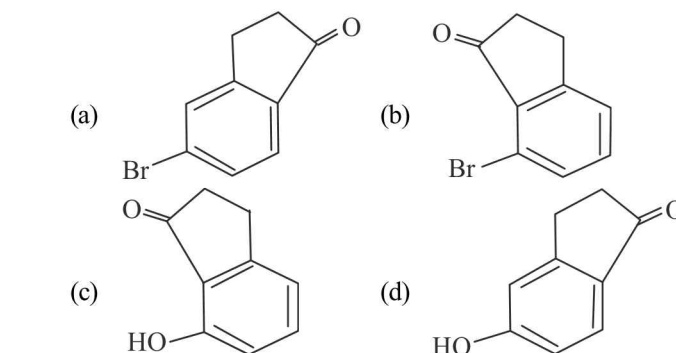
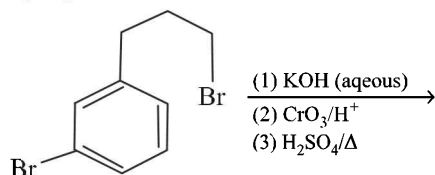
25. Which compound would give 5 - keto - 2 - methylhexanal upon ozonolysis ?



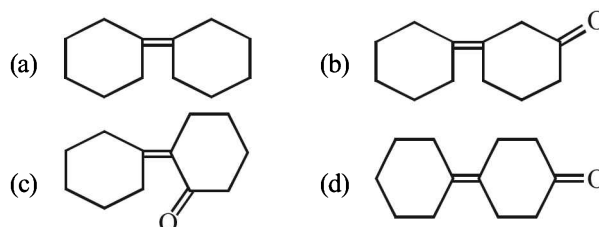
26. Aldehydes and ketones will not form crystalline derivatives with :

- (a) sodium bisulphite
 (b) phenylhydrazine
 (c) semicarbazide hydrochloride
 (d) dihydrogen sodium phosphate.

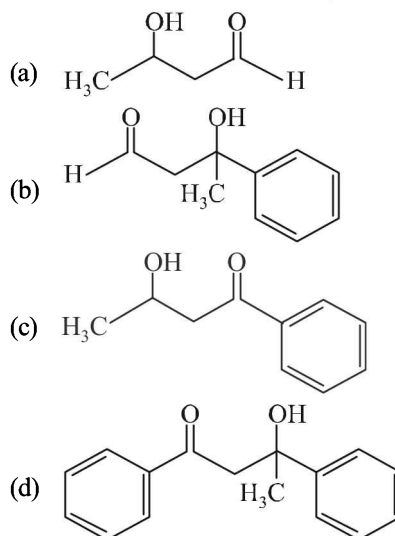
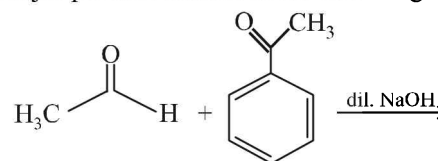
27. The major product of the following reaction is:



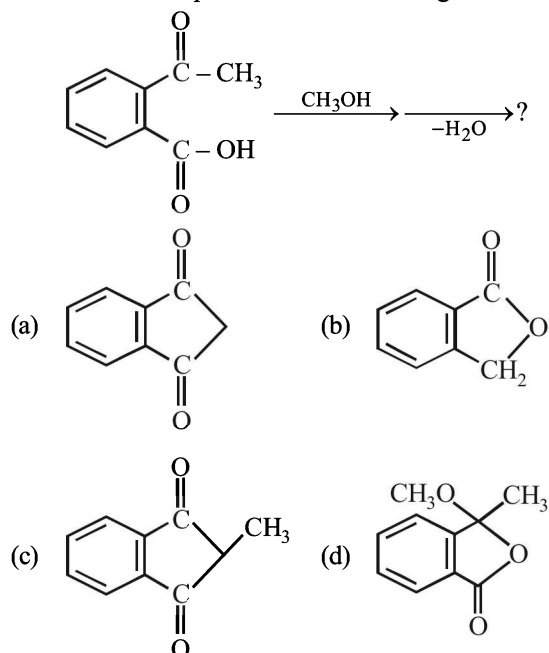
Product is :



29. The major product formed in the following reaction is:



30. What is the final product of the following reaction ?



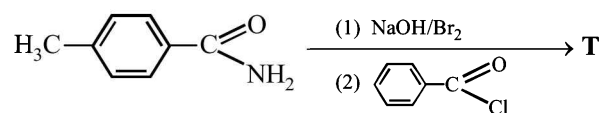
31. An ester (A) with molecular formula, $\text{C}_9\text{H}_{10}\text{O}_2$ was treated with excess of CH_3MgBr and the complex so formed was treated with H_2SO_4 to give an olefin (B). Ozonolysis of (B) gave a ketone with molecular formula $\text{C}_8\text{H}_8\text{O}$ which shows positive iodoform test. The structure of (A) is :

- (a) $\text{C}_6\text{H}_5\text{COOC}_2\text{H}_5$
 (b) $\text{C}_2\text{H}_5\text{COOC}_6\text{H}_5$
 (c) $\text{H}_3\text{COCH}_2\text{COC}_6\text{H}_5$
 (d) $p\text{-H}_3\text{CO}-\text{C}_6\text{H}_4-\text{COCH}_3$

32. Which of the following is an example of aldol condensation?

- (a) $2\text{CH}_3\text{COCH}_3 \xrightarrow{\text{dil NaOH}} \text{CH}_3\text{C}(\text{OH})(\text{CH}_3)\text{CH}_2\text{COCH}_3$
 (b) $2\text{HCHO} \xrightarrow{\text{dil NaOH}} \text{CH}_3\text{OH}$
 (c) $\text{C}_6\text{H}_5\text{CHO} + \text{HCHO} \xrightarrow{\text{dil NaOH}} \text{C}_6\text{H}_5\text{CH}_2\text{OH}$
 (d) None of the above

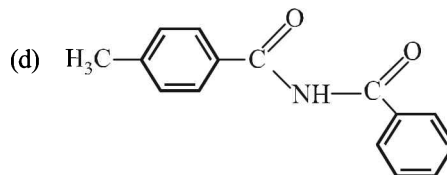
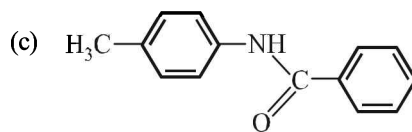
33. In the reaction



the structure of the product T is :

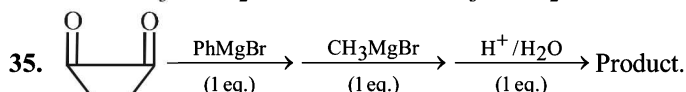
- (a)

 (b)



34. When $\text{CH}_2=\text{CH}-\text{COOH}$ is reduced with LiAlH_4 , the compound obtained will be

- (a) $\text{CH}_2=\text{CH}-\text{CH}_2\text{OH}$ (b) $\text{CH}_3-\text{CH}_2-\text{CH}_2\text{OH}$
 (c) $\text{CH}_3-\text{CH}_2-\text{CHO}$ (d) $\text{CH}_3-\text{CH}_2-\text{COOH}$

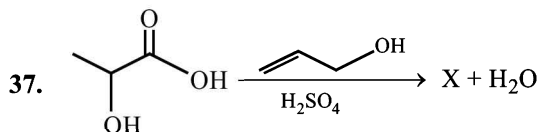


The product formed in the reaction is –

- (a)
 (b) $\text{Me}-\text{C}(=\text{O})-\text{C}(=\text{O})-\text{Ph}$
 (c) $\text{Me}-\text{C}(\text{OH})(\text{Ph})-\text{COOH}$
 (d) $\text{Ph}-\text{C}(=\text{O})-\text{C}(=\text{O})-\text{O}-\text{Me}$

36. The strongest acid among the following is –

- (a) Salicylic acid (b) *m*-hydroxybenzoic acid
 (c) *p*-hydroxybenzoic acid (d) Benzoic acid



Product (X) of the reaction is–

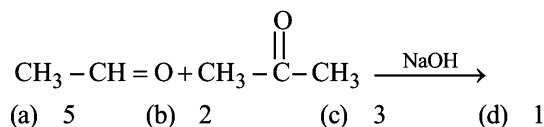
- (a)
 (b)

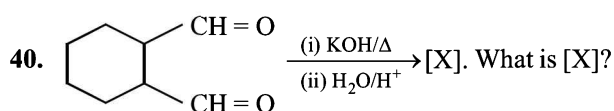
 (c)
 (d)

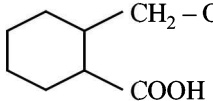
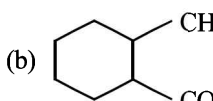
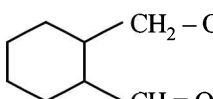
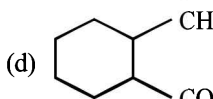
38. $\text{CH}_3\text{COOH} \xrightarrow{\text{A}} \text{CH}_3\text{COCl}$. What is A ?

- (a) PCl_5 (b) Cl_2 (c) HCl (d) COCl_2

39. How many cross aldol products (without counting stereoisomers) are produced in the following reaction?



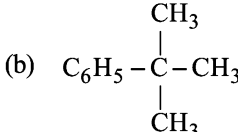
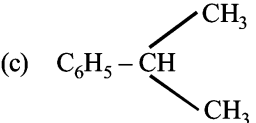


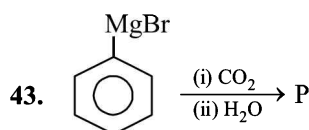
- (a)  (b) 
 (c)  (d) 

41. Order of hydrolysis for the following :

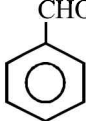
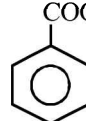
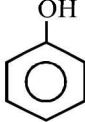
- (I) RCOCl (II) RCOOR
 (III) RCONH_2 (IV) $(\text{RCO})_2\text{O}$
 (a) $\text{I} > \text{IV} > \text{II} > \text{III}$ (b) $\text{I} > \text{II} > \text{III} > \text{IV}$
 (c) $\text{I} > \text{III} > \text{II} > \text{IV}$ (d) $\text{IV} > \text{III} > \text{II} > \text{I}$

42. Which of the following compound will not give benzoic acid on oxidation with $\text{KMnO}_4/\text{OH}^-/\Delta$ -

- (a) $\text{C}_6\text{H}_5-\text{CH}_3$ (b) 
 (c)  (d) $\text{C}_6\text{H}_5-\text{CH}_2-\text{COOH}$



In the reaction, product P is :

- (a)  (b) 
 (c)  (d) $\text{C}_6\text{H}_5-\text{C}(=\text{O})-\text{C}_6\text{H}_5$

44. The reactivity of carbonyl compounds towards nucleophilic addition reaction increases by which factor?

- (a) Magnitude of positive charge on carbonyl carbon atom
 (b) Steric hindrance
 (c) Presence of electron withdrawing groups surrounding carbonyl group
 (d) Both (a) and (c)

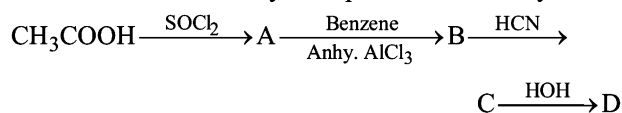
45. The increasing order of the rate of HCN addition to compound A - D is

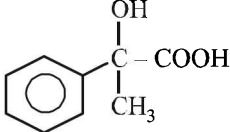
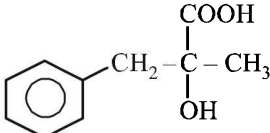
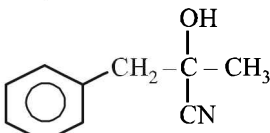
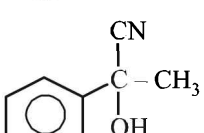
- (A) HCHO (B) CH_3COCH_3
 (C) PhCOCH_3 (D) PhCOPh
 (a) $\text{D} < \text{C} < \text{B} < \text{A}$ (b) $\text{C} < \text{D} < \text{B} < \text{A}$
 (c) $\text{A} < \text{B} < \text{C} < \text{D}$ (d) $\text{D} < \text{B} < \text{C} < \text{A}$



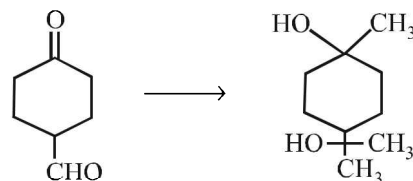
Skill Based MCQs

46. In a set of reactions acid yield a product D. Identify D :

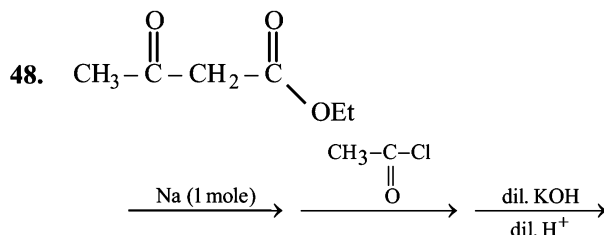


- (a) 
 (b) 
 (c) 
 (d) 

47. The correct sequence of reagents for the following conversion will be :

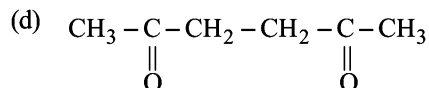
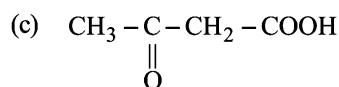
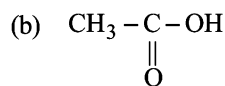


- (a) $[\text{Ag}(\text{NH}_3)_2]^+ \text{OH}^-$, $\text{H}^+/\text{CH}_3\text{OH}$, CH_3MgBr
 (b) CH_3MgBr , $\text{H}^+/\text{CH}_3\text{OH}$, $[\text{Ag}(\text{NH}_3)_2]^+ \text{OH}^-$
 (c) CH_3MgBr , $[\text{Ag}(\text{NH}_3)_2]^+ \text{OH}^-$, $\text{H}^+/\text{CH}_3\text{OH}$
 (d) $[\text{Ag}(\text{NH}_3)_2]^+ \text{OH}^-$, CH_3MgBr , $\text{H}^+/\text{CH}_3\text{OH}$

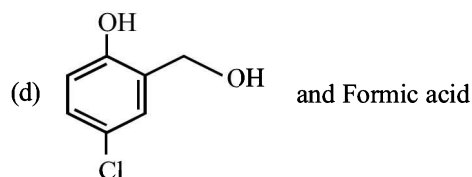
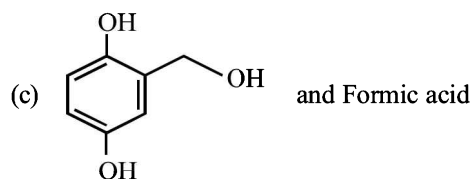
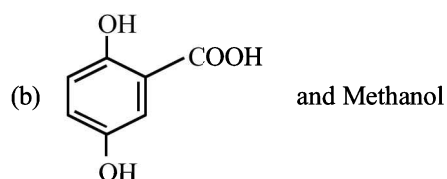
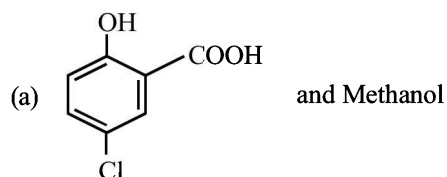
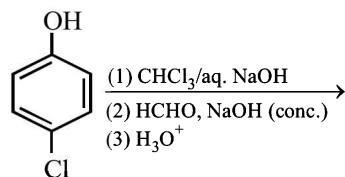


The final product is -

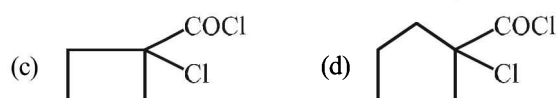
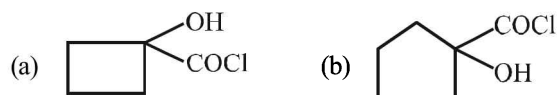
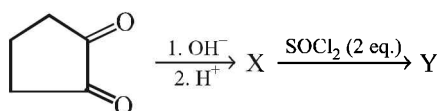
- (a) $\text{CH}_3-\text{C}(=\text{O})-\text{CH}_2-\text{C}(=\text{O})-\text{CH}_3$



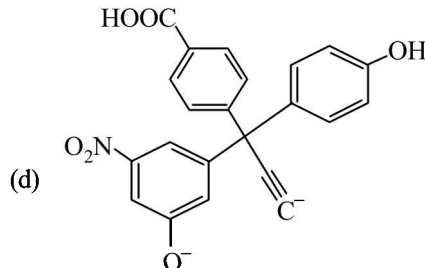
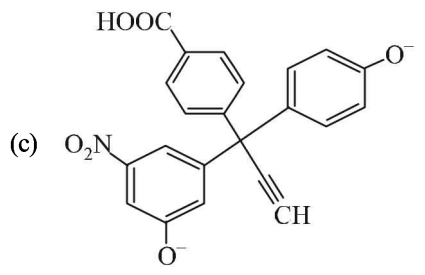
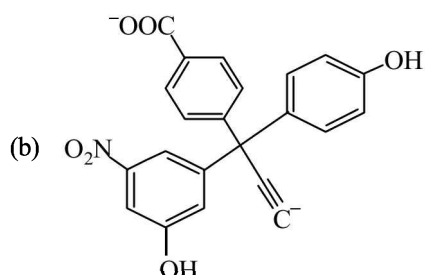
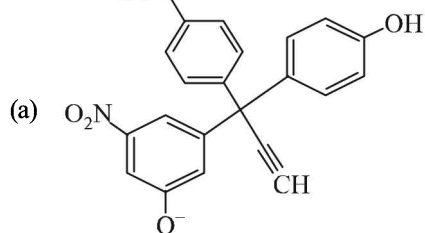
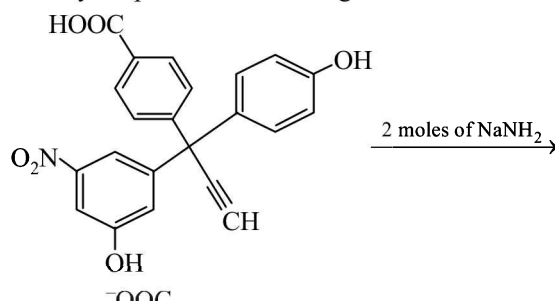
49. The major products of the following reaction are :



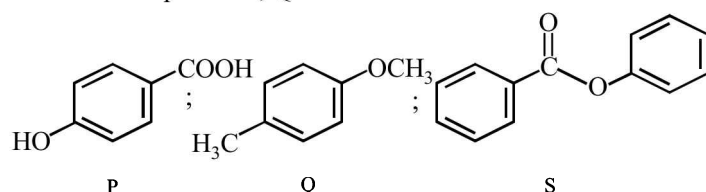
50. In the following sequence of reaction find the product Y:



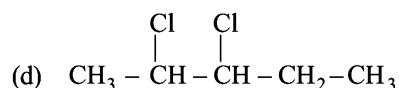
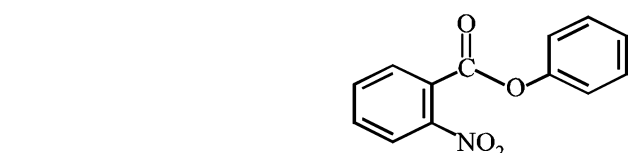
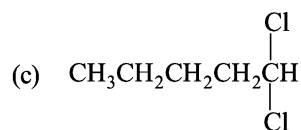
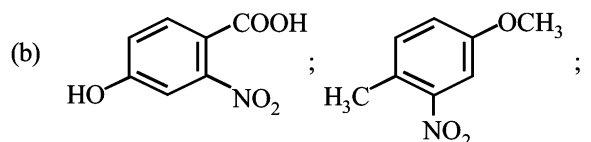
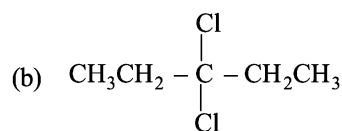
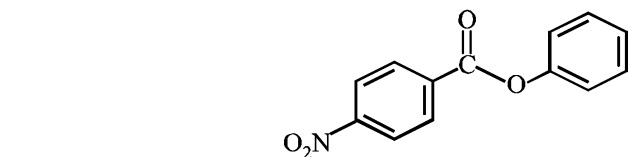
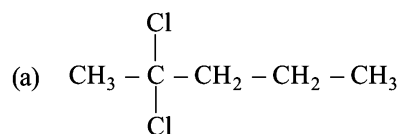
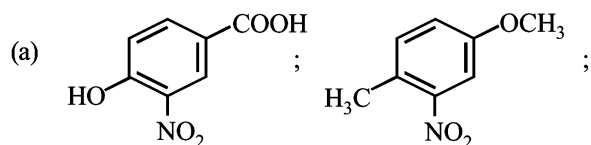
51. Identify the product in following reaction



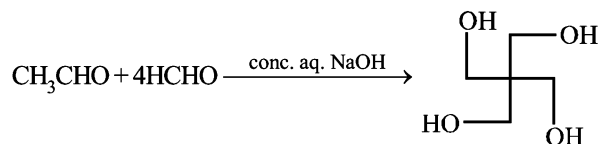
52. The compounds P, Q and S



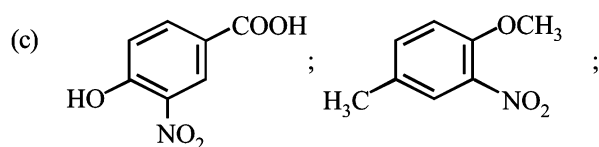
were separately subjected to nitration using $\text{HNO}_3/\text{H}_2\text{SO}_4$ mixture. The major product formed in each case respectively, is :



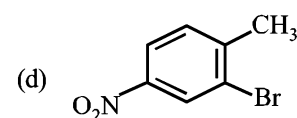
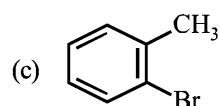
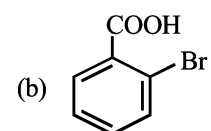
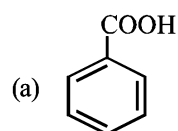
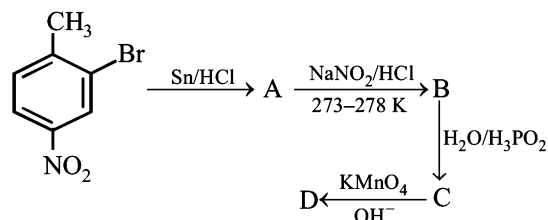
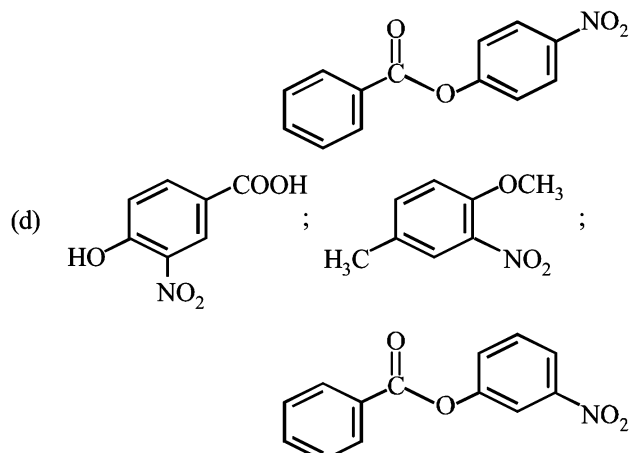
54. The number of aldol reaction(s) that occurs in the given transformation is :



(a) 1 (b) 2 (c) 3 (d) 4



55. Identify the product (D) in the following sequence of reactions.



53. A compound A ($\text{C}_5\text{H}_{10}\text{Cl}_2$) on hydrolysis gives $\text{C}_5\text{H}_{10}\text{O}$ which reacts with NH_2OH and forms iodoform but does not give Fehling test. A is :

ANSWER KEY

Conceptual MCQs

1	(a)	3	(c)	5	(c)	7	(b)	9	(a)	11	(c)	13	(a)	15	(d)	17	(a)	19	(a)
2	(c)	4	(d)	6	(c)	8	(b)	10	(b)	12	(d)	14	(c)	16	(b)	18	(d)	20	(c)

Application Based MCQs

21	(d)	24	(b)	27	(a)	30	(d)	33	(c)	36	(a)	39	(b)	42	(b)	45	(a)		
22	(d)	25	(d)	28	(c)	31	(a)	34	(a)	37	(b)	40	(a)	43	(b)				
23	(c)	26	(d)	29	(c)	32	(a)	35	(c)	38	(a)	41	(a)	44	(d)				

Skill Based MCQs

46	(a)	47	(a)	48	(a)	49	(d)	50	(c)	51	(a)	52	(c)	53	(a)	54	(c)	55	(b)
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