

SECTION- I (ONE OR MORE THAN ONE)

Each question has FOUR options for correct answer(s). ONE OR MORE THAN ONE of these four option(s) is (are) correct option(s).

For each question, choose the correct option(s) to answer the question.

Answer to each question will be evaluated according to the following marking scheme:

Full Marks: +4 If only (all) the correct option(s) is (are) chosen.

Partial Marks: +3 If all the four options are correct but ONLY three options are chosen.

Partial Marks: +2 If three or more options are correct but ONLY two options are chosen, both of which are correct

options.

Partial Marks: +1 If two or more options are correct but ONLY one option is chosen and it is a correct option.

Zero Marks: 0 If none of the options is chosen (i.e. the question is unanswered).

Negative Marks: -2 In all other cases.

1. The **IUPAC** name(s) of the following compound is (are):

- A) 1-chloro-2-bromocyclohexa-1,3-diene
- B) 1-bromo-2-chlorocyclohexa-1,3-diene
- C) 2-bromo-3-chlorocyclohexa-1,3-diene
- D) 2-chloro-1-bromocyclohexa-1,3-diene

2. Which of the following is NOT the chain isomer of

3. Absolute stereochemistry of the given compound is?

 The reactant in which diastereomeric mixture is not obtained as a major product on reaction with Br₂ / CCl₄ is

$$\begin{array}{c} \text{CH}_3 \\ \text{A)} \\ \\ \text{B)} \\ \\ \text{CH}_3 \\ \\ \text{CH}_3 \\ \\ \text{CH}_3 \\ \\ \text{CH}_2 \\ \\ \text{D)} \\ \\ \text{COOH} \\ \\ \text{COOH} \\ \\ \text{COOH} \\ \\ \text{CI}_2 \text{H}_10 \text{O}_2) \\ \\ \text{Optical active} \\ \\ \text{optical inactive} \\ \\ \text{[R]} \\ \end{array}$$

P gives CO2 gas with NaHCO3.

5.

Correct statement(s) is(are) for the above reactions:

- A) Total monochloro products obtained by Cl_2/hv of compound (Q) including stereoisomers are 7.
- B) IUPAC name of **(P)** is 3-(1-Ethynylprop-2-enyl) benzene carboxylic acid.
- C) Compound (Q) gives effervescence of CO2 with sodium bicarbonate
- D) One monochloro product of \mathbf{Q} is achiral

The energy profile for the two possible mechanisms of the following reaction, 6.

$$R_2$$
 $\xrightarrow{R_1}$ Br $+$ HO \longrightarrow R_2 $\xrightarrow{R_1}$ OH $+$ Br

Are given below





Consider the following substrates (I to IV) for the above reaction

I.
$$R_1 = R_2 = R_3 = H$$

II.
$$R_1 = C_2H_5$$
; $R_2 = R_3 = H$

III.
$$R_1 = C_2H_5$$
; $R_2 = CH_3$; $R_3 = H$ IV. $R_1 = C_2H_5$; $R_2 = R_3 = CH_3$

IV.
$$R_1 = C_2H_4$$
; $R_2 = R_3 = CH_4$

Choose the CORRECT statement(s).

- A) The reaction mechanism changes from the one shown in figure Y to that in figure X, as one changes the substrate from I to IV.
- B) Considering the mechanism corresponding to figure X for substrates I to IV, the correct order of ΔE is : $\Delta E_{II} > \Delta E_{III} > \Delta E_{III} \Delta E_{IV}$.
- C) The reaction for substrate I follows 2nd order kinetics and that for IV follows 1st order kinetics.
- D) If OH is taken in large excess, reaction may not follows mechanism corresponding to energy profile in figure X.
- 7. Arrange the following compounds in increasing order of C-OH bond length

Methanol, Phenol, p-methoxyphenol

- A) p-methoxyphenol< phenol < methanol
- B) Phenol< p-methoxyphenol< methanol
- C) Methanol< p-methoxyphenol< phenol
- D) Methanol < p-methoxyphenol

9,

The intermediate(s) formed in the above reaction is/ are:

SECTION- II (NUMERICAL VALUE)

The answer to each question is a NUMERICAL VALUE

For each question, enter the correct numerical value (in decimal notation, truncated/rounded off to the second decimal place; e.g. 6.25, 7.00, -0.33, -.30, 30.27, -127.30) designated to enter the answer.

Answer to each question will be evaluated according to the following marking scheme:

+3 If ONLY the correct numerical value is entered as answer. Full Marks:

Zera Marks: O in all other cases.

- 10. The degree of unsaturation (double bond equivalent) for a compound with molecular formula C14H12O2 is
- 11. Total number of constitutional isomers (only position isomers) possible for trimethyl cyclohexane is
- Among the following the total number of set(s) of diastereomeric pair(s) is? 12.

NH2

and

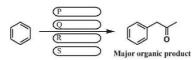
NH₂

Ph

13. Find the number of unsaturated products formed in given reaction

14. Choose the appropriate reagent sequence from the given reagents pool to complete the synthetic conversation. Make sure to use strictly four reagents, not more not less. Note: you can simply mention the numerical values corresponding to the reagents in the boxes P, Q, R, S, respectively and also consider major organic product in each step.

What is the value of [P + Q + R - S]



- 1. CH₃CH₂CH₂Cl, AlCl₃
- 5. CH₃Cl, AlCl₃
- 9. CH₃CH₂Cl, AlCl₃

- 2. NBS, ROOR, hv
- 6. H₂O₂, NaOH, H₂O
- 10. PCC

3. BH₃·THF

7. HBr

4. HC≡C

- 8. H₂O, H₂SO₄, HgSO₄
- 15. If [X] is the number of different compounds produced on monochlorination of cis-1,2-dimethyl cyclopentane then find $\left\lceil \frac{X}{2} \right\rceil$? (Including stereoisomers).

SECTION- III (MATRIX MATCH WITH NUMERICAL)

Each question has TWO (02) matching lists: LIST-I and LIST-II.

FOUR options are given representing matching of elements from LIST-II and LIST-II.

For each question, choose the option corresponding to the correct matching.

For each question, choose the option corresponding to the correct matching.

Full Marks : +3 If ONLY the option corresponding to the correct matching is chosen.

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered).

Negative Marks : -1 in all other cases.

16. Match the column

	List-I (Compound)		List-II (IUPAC names)
P) (CI Br	1)	1-Bromo-4-chloro cyclohexene
Q) (CI	2)	4-Bromo-1-chloro cyclohexene
R) c		3)	1-Bromo-2-chloro cyclohexene
S) B	CI	4)	6-Bromo-1-chloro cyclohexene

Note: Correctly match P, Q, R, S with 1, 2, 3 & 4 and give your answer in integer values

If

P	Q	R	S
4	1	2	3

Then your answer is [4123]

ANSWERS

1	2	3	4	5	6	7	8	9	10
С	D	D	A	ABCD	ACD	В	AC	A	9
11	12	13	14	15	16	17	18	19	20
6	3	3	3	6	3412				