DPP No. 10

Total Marks: 49

Max. Time: 50 min.

Topic: General Organic Chemistry

(A) 3

(B) 4

Type of Questions M.M., Min. (3 marks, 3 min.) Single choice Objective ('-1' negative marking) Q.1 to Q.15 [45, 45] Subjective Questions ('-1' negative marking) Q.16 (4 marks, 5 min.) [4, 5] 1. How many lone pairs of electrons are there in the given compound? (A) 4 (B) 2 (C) 8(D) 6 2. How many lone pairs of electrons are there in the given compound? (A) 4(B) 3 (C) 8(D) 6 3. What is the hybridisation of positively charged nitrogen atom? (A) sp (B) sp² (D) None of these (C) sp^3 4. How many sp² hybridised atoms are there in the given cation? (B) 9 (A) 8 (C) 10 (D) 12 5. How many lone pairs are present in the given cation? (A) 3 (B) 4 (C) 5 (D) 6 6. How many N atoms are sp² hybridised in the given cation?

(C) 5

(D) 6

7. In which of the following molecules is the nitrogen atom sp² hybridised?

(A)
$$HO_2C$$
 N
 CO_2H
 CO_2H
 N
 OH
 OH
 OH

8. Geometry around how many carbon atoms is tetrahedral in the given structure?

9. How many sp³ hybridised carbon atoms are there in the given anti-cancer compound (podophyllotoxin) ?

(A) 4

the anti-cancer compound podophyllotoxin

- (A) 6 (B) 7 (C) 9 (D) 8
- 10. Bond order of C–C bond in benzene is:

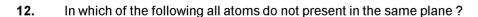
 (A) 1 (B) 2 (C) 1.5 (D) two of above
- 11. Which of the following is correct three dimentional representation of CH_{4} ?

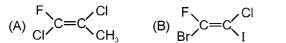
$$(A) H - C - H$$

$$(B) H - H$$

$$(C) H - H$$

$$(D) + H$$

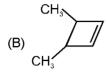




(B)
$$F C = C C$$

13. In which of the following all carbon atoms are present in the same plane?







14. In which of the following are all C atoms linearly arranged?

(A)
$$CH_2 = CH - C = CH$$
 (B) $CH = C - CH_2 - CH_3$ (C) $CH_3 - C = C - CH_3$ (D) $CH_3 - C = C - CH_3$

15. In
$$\frac{CH_3}{CH_3}C = C = C \frac{CH_3}{CH_3}$$
 How many carbon atoms are linearly arranged?

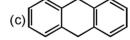
(A) 1

- (B) 8
- (C) 3
- (D) 7

16. A maximum of how many (i) atoms (ii) carbon atoms must lie in same plane in each of the following molecule?









No. of sp²–sp²

 σ -bonds

No. of sp2-sp

 $\sigma\text{-bonds}$

Answer Key

DPP No. #10

1. (D) 2.

(B)

3. (C) 4. (C) 5. (B)

6. (C) 7. (C)

(C) 8.

9. (C) 10. (C)

(C)

11. (C) 12.

(A)

13.

14.

(D)

15.

16.

(i) 12, (ii) 6.

(b)

(f)

(i) 12, (ii) 8.

(c)

(C)

(i) 12, (ii) 8.

(d).

(i) 6, (ii) 4.

(e)

(a)

(i) 14, (ii) 8.

(i) 14, (ii) 9.